

## Liposomes

Advancements and innovation in the manufacturing process

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## **Liposomes: Advancements and innovation in the manufacturing process**

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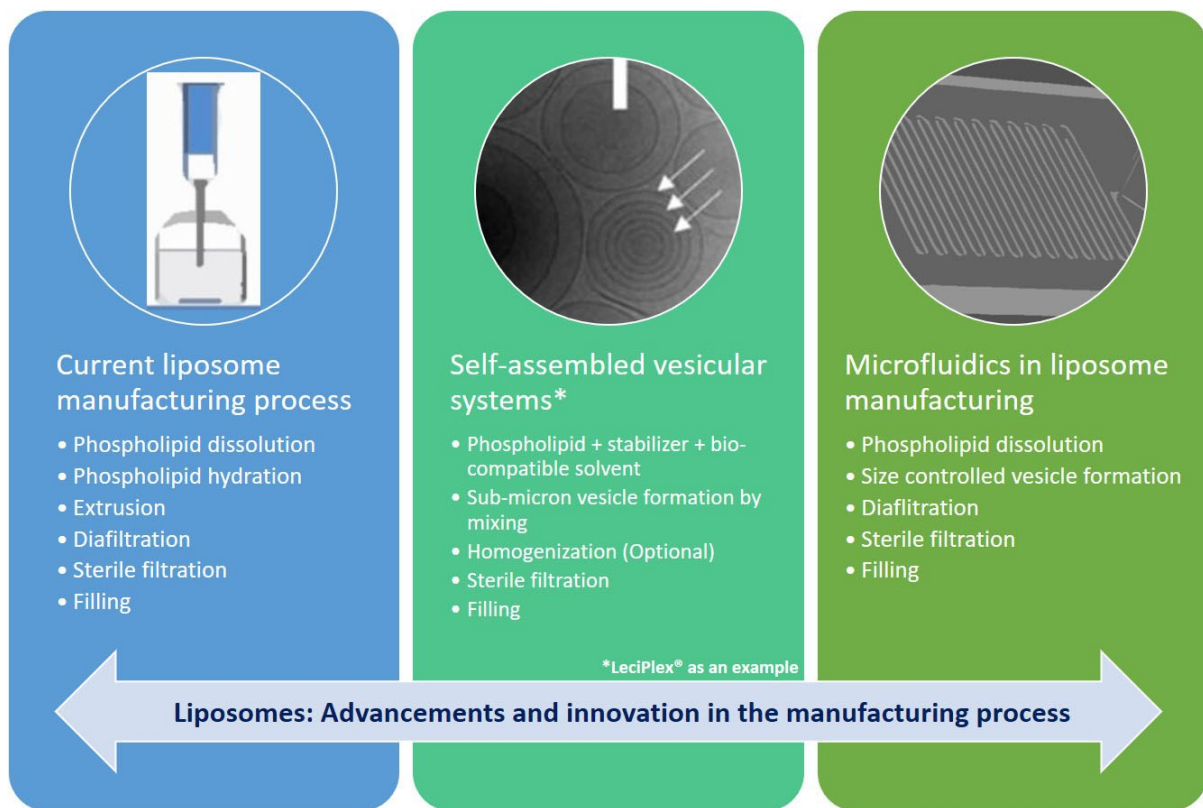
## 1    **Highlights**

- 2    • Liposomes and their related constructs offer unique advantages in terms of drug and  
3    vaccine delivery.
- 4    • However, current processes used for the manufacture of liposomes present a range of  
5    challenges, driving up cost, and limiting production.
- 6    • New production methods can address these issues and support the cost-effective  
7    manufacture of current liposomal systems and facilitate the development of new  
8    liposomal products.

## 1   **Abstract**

2   Liposomes are well recognised as effective drug delivery systems, with a range of products  
3   approved, including follow on generic products. Current manufacturing processes used to  
4   produce liposomes are generally complex multi-batch processes. Furthermore, liposome  
5   preparation processes adopted in the laboratory setting do not offer easy translation to large  
6   scale production, which may delay the development and adoption of new liposomal systems.  
7   To promote advancement and innovation in liposome manufacturing processes this review  
8   considers the range of manufacturing processes available for liposomes, from laboratory scale  
9   and scale up, through to large-scale manufacture and evaluates their advantages and  
10   limitations. The regulatory considerations associated with the manufacture of liposomes is  
11   also discussed. New innovations that support leaner scalable technologies for liposome  
12   fabrication are outlined including self-assembling liposome systems and microfluidic  
13   production. The critical process attributes that impact on the liposome product attributes are  
14   outlined to support potential wider adoption of these innovations.

## 1 Graphical abstract



2

1    **Keywords**

2    Liposomes; Lipid nanoparticulate; Microfluidics; Leciplex; Phospholipids; cationic liposomes;  
3    nanomedicine; monoclonal antibody; oligonucleotide; targeted delivery

4

## 1    **Abbreviations**

- 2    Chol: Cholesterol
- 3    cryoTEM: cryo transmission electron microscopy
- 4    Da: Daltons
- 5    DLin-MC3-DMA: (6Z,9Z,28Z,31Z)-heptatriaconta-6,9,28,31-tetren-19-yl-4-
- 6    (dimethylamino)butanoate
- 7    DMG-PEG2000: (R)-2,3-bis(tetradecyloxy)propyl 1-(methoxypoly(ethylene
- 8    glycol)20000)propyl carbamate
- 9    DOPE: 1,2-dioleoyl-sn-glycero-3-phosphoethanolamine
- 10   DOTAP: 1,2-dioleoyl-3-trimethylammonium-propane (chloride salt)
- 11   DPPC: 1,2-Dipalmitoyl-3-sn-phosphatidylcholine
- 12   DSPC: 1,2-distearoyl-sn-glycero-3-phosphocholine
- 13   EMA: European medicines agency
- 14   FDA: Food and drug administration
- 15   FRR: flow rate ratio
- 16   FVR: flow velocity ratio
- 17   GMP: Good manufacturing practice
- 18   GUVs: Giant unilamellar vesicles
- 19   HGL: High gravity level
- 20   ICH: International Council for Harmonisation
- 21   IPA: iso-propyl alcohol
- 22   IPQC: In-process quality control
- 23   LFH: Lipid film hydration
- 24   LNPs: Lipid nanoparticles

- 1 Log P: Partition co-efficient
- 2 LUVs: Large Unilamellar vesicles
- 3 MC: microchannel
- 4 MLVs: Multilamellar vesicles
- 5 MVVs: Multivesicular vesicles
- 6 OLVs: Oligo lamellar vesicles
- 7 PBS: Phosphate buffer saline
- 8 PC: Phosphatidyl choline
- 9 PDI: Polydispersity index
- 10 PTFE: polytetrafluoroethylene
- 11 PVA: polyvinyl alcohol
- 12 RNAi: Ribose nucleic acid interference
- 13 siRNA: small interfering ribose nucleic acid
- 14 SUVs: Small unilamellar vesicles
- 15 TBA: tert-butyl alcohol
- 16 TFF: tangential flow filtration
- 17 ULVs: unilamellar vesicles
- 18 w/o/w emulsion: Water-in-oil-in-water emulsion
- 19 w/o: water-in-oil
- 20



1	<b>Contents</b>	
2	Highlights .....	2
3	Abstract.....	3
4	Graphical abstract .....	4
5	Keywords.....	5
6	Abbreviations.....	6
7	<b>1. Introduction and overview of application of liposomes.....</b>	<b>9</b>
8	<b>2. Manufacturing of liposomes .....</b>	<b>13</b>
9	2.1. Laboratory scale manufacturing .....	13
10	2.1.1. Solvent evaporation.....	15
11	2.1.2. Solvent dispersion.....	16
12	2.1.3. Reverse phase evaporation.....	16
13	2.1.4. Size manipulation.....	17
14	2.1.5. Final liposomal drug product .....	17
15	2.2. Industrial manufacturing and scale up considerations.....	19
16	2.3. Potential for innovation in large scale liposome manufacturing.....	23
17	3. Regulatory overview of liposomes.....	25
18	4. Advances in scalable technologies for liposome fabrication .....	30
19	4.1. Self-assembled vesicular drug delivery system .....	31
20	4.1.1. Heating methods.....	31
21	4.1.2. Nanoprecipitation and Ionic interaction.....	32
22	4.1.3. Solvent exchange .....	33
23	4.1.4. High Shear .....	34
24	4.1.5. Emulsification and solvent evaporation .....	35
25	4.1.6. Packed bed-based reactors.....	36
26	4.1.7. Gel assisted self-assembly .....	38
27	4.1.8. Spray drying and fluid bed drying .....	39
28	4.1.10. Freeze drying.....	40
29	4.1.11. Supercritical fluid techniques.....	41
30	4.2. Microfluidics.....	45
31	4.2.1. Microfluidic cartridge design .....	46
32	4.2.2. Microfluidic material and production parameter considerations .....	49
33	<b>5. Conclusion .....</b>	<b>56</b>
34	<b>6. References .....</b>	<b>58</b>

35

36

## 1. Introduction and overview of application of liposomes

Lipid-based nanomedicines are used to 1) protect drugs from degradation *in vivo*, 2) control drug release, 3) modify biodistribution, 4) target drug delivery to the site of disease and 5) enhance solubility and bioavailability. Lipid based delivery systems are also effective as vaccine adjuvants through their ability to protect and deliver antigens (peptide, protein and nucleic acid systems) to the antigen presenting cells and stimulating protective immune responses. Suitable engineering of nanomedicines in terms of their composition, particle size, and surface charge can aid in achieving spatial and temporal delivery of drugs. This applies to the delivery of traditional small molecules and to lipid-based nanoparticles used to deliver nucleic acid-based drugs; patisiran (Onpattro®; Alnylam), approved by the FDA and the EMA in 2018, is the first siRNA-based drug approval. Patisiran is indicated for polyneuropathy of hereditary transthyretin-mediated amyloidosis. Onpattro® contains 2.0 mg/mL of patisiran (a double stranded siRNA which is the active substance) incorporated into lipid nanoparticles. The nanoparticles are 60 – 100 nm in size with a near neutral surface charge at biological pH. The nanoparticles are formed during production as a result of the lipids associating with the siRNA. The nanoparticles are built from four lipids. DSPC is used with cholesterol to support the formation and stability of the lipid nanoparticles. An ionisable lipid (DLin-MC3-DMA) is incorporated which electrostatically interacts with the siRNA and promotes high drug loading and a pegylated lipid DMG-PEG2000 is incorporated to improve the stability of the formed LNPs [1]. Of these lipid-based nanomedicines, liposomes are generally the most well-established systems.

Liposomes are lipid based spherical shaped vesicular systems, in which a lipophilic bilayer is sandwiched between two hydrophilic layers. The versatility and advantages of liposomes as a drug delivery system for small molecules, peptides, gene, and monoclonal antibodies is well studied and acknowledged in the peer-reviewed scientific literature [2-6]. Liposomes fall into the general category of nanomedicines and play a key role in many diverse areas of health and have found an application in the treatment of patients suffering from cardiovascular disease, neurodegenerative disease, diabetes, cancer and inflammation. Parenteral delivery offers the advantage of bypassing first pass metabolism, poor gastrointestinal permeability and gastrointestinal side effects (a problem common to oral delivery of drugs) and parenteral

1 administration provides an opportunity for targeted delivery of drugs resulting in higher  
2 bioavailability and reduced off target side effects.

3 A myriad of reports exist in the literature on application of liposomes to deliver drugs  
4 [7-10] and genes [11-14] parenterally. Phospholipids being biodegradable and biocompatible  
5 and bearing resemblance to the lipids present in cellular membranes, are widely explored for  
6 their drug delivery potential associated with their assemblies to organise structure (**Fig. 1**).  
7 Liposomes on account of being formed due to self-assembly possess a thermodynamic  
8 advantage. They have been used for cancer treatment to improve tumour targeting and  
9 reduce off-target toxicity (e.g. Doxil®) and to treat patients with severe infections or  
10 immunocompromised conditions (e.g. AmBisome®). Whilst access to these advanced  
11 treatments can be limited (mainly due to their cost), in 2015, the global liposomal doxorubicin  
12 market alone was valued at USD 814.6 million [15]. Furthermore, there are many other  
13 marketed oncology nanomedicines using lipid-based nanotechnology (e.g. DaunoXome®,  
14 Myocet®, DepoCyt®, Marqibo® and Onivyde®) and more recently Onpattro®, the first FDA-  
15 approved RNAi therapeutic. Indeed, the nanomedicines market is recognised as a high risk,  
16 high return market and has enjoyed unprecedented growth over the last five years.

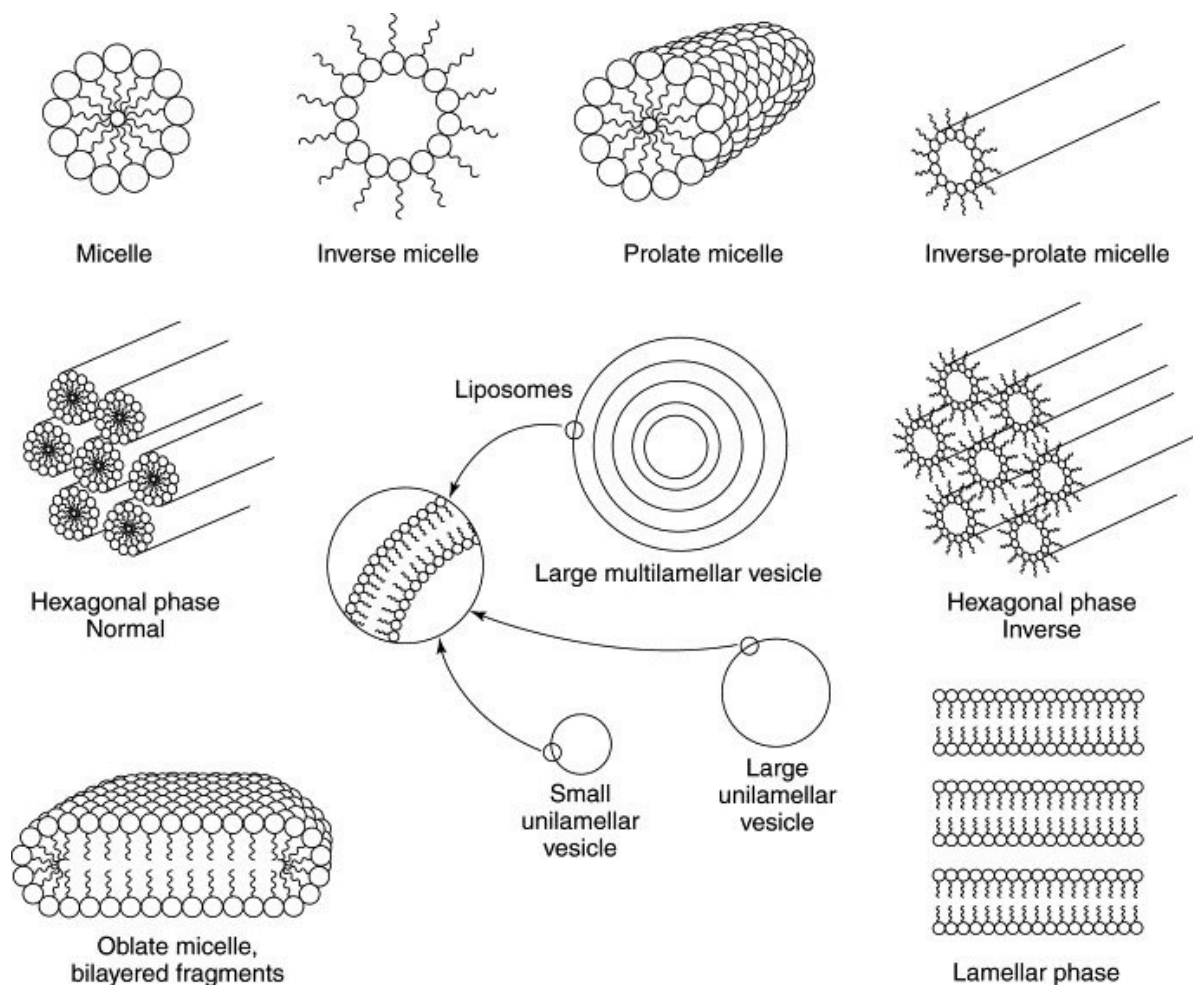


Fig. 1. Schematic showing the different phases of a phospholipid molecule. The phospholipid molecules in an aqueous environment, under right conditions, forms a bilayered lamella spherical in shape called liposomes. Depending on the types and forms, liposomes can be called a LUVs, SUVs, and MLVs. Reprinted from Trends in Biotechnology, 16/7, Dan D Lasic, Novel applications of liposomes, 307-321, Copyright (2020), with permission from Elsevier [16].

The global nanomedicine market was valued at USD 135 billion in 2015 and it is anticipated that this will reach USD 350.8 billion by 2025 [17]. However, using traditional manufacturing methods, the scale-up production of nanomedicines presents a significant challenge to their clinical development and the cost of commercial manufacture is a recognised barrier to their translation from bench to bedside. Despite their widespread research, it is well recognised that the current processes used for manufacturing of liposomes suffers from many severe problems, including: i) multi-step batch processes; ii) the need for particle size reduction (often involving specialized tools and equipment such as extrusion and high-pressure homogenization) and iii) limited batch sizes. This drive cost upwards, limits production and hinders development of liposomes. This was exemplified by the global shortage of Doxil® due to closure of a sterile injectables production site due to manufacturing

1 challenges as the principle reason. Global shortages of this anti-cancer treatment lasted for  
2 more than two years [18]. Hence, it is important to identify ways of making the liposomal  
3 manufacture process leaner and identifying ways to make this drug delivery option more  
4 attractive from an industrial point of view.

## 2. Manufacturing of liposomes

### 2.1. Laboratory scale manufacturing

Numerous reviews and research articles have been published elsewhere on the composition, preparation, and characterization of liposomes and readers are requested to refer them for more in-depth understanding [19-30]. Almost all the techniques involve dissolution of phospholipids in an organic solvent followed by removal of the organic solvent, later in the process. This prior dissolution followed by removal of organic solvent is important for the formation of liposomes. The building blocks of liposomes are phospholipids and/or cholesterol. The critical micelle concentration of most commonly used phospholipids is in the nanomolar range and the concentration of phospholipids used for liposomes manufacturing is much above the critical micelle concentration. This along with the three-dimensional cylinder like shape of each phospholipid (**Fig. 2**) leads to formation of liposomes along with lipid aggregates when phospholipids, as such, are exposed to an aqueous environment. In order to make uniform liposomal dispersions, it is important to make thin lipid sheets before exposing it to an aqueous phase or introduce the organic phospholipid solution in a controlled manner in an aqueous environment for the formation of liposomes. This is why all the reported techniques of liposome manufacturing i.e. solvent evaporation, solvent dispersion/antisolvent addition, or detergent removal focus on first disaggregating the phospholipids into individual phospholipid molecules followed by exposure to aqueous environment to enable formation of different types of liposomes viz. MLVs, SUVs, GUVs, OLVs, MVVs (**Table 1**) [31, 32]. The detergent removal technique is not discussed in this , but interested reader can be referred to other literature sources [33, 34]. A special mention of the reverse phase evaporation method is discussed as it is the preferred laboratory technique for obtaining high entrapment efficiency of hydrophilic drugs [35].

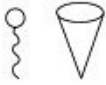
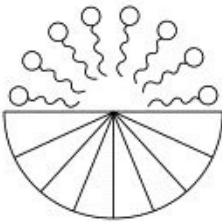

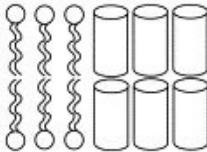

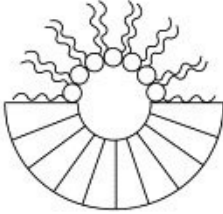
Species	Shape	Organization	Phase
Soaps Detergents Lysophospholipids	 Inverted cone	 Micelles	Isotropic Hexagonal 1
Phosphatidylcholine Phosphatidylserine Phosphatidylinositol Sphingomyelin Dicetylphosphate DODAC	 Cylinder	 Bilayer	Lamellar (Cubic)
Phosphatidylethanolamine Phosphatidic acid Cholesterol Cardiolipin Lipid A	 Cone	 Reverse micelles	Hexagonal 2

Fig. 2. Different molecular shapes of a surfactant/phospholipid like molecules. Most of the commonly used phospholipids have a cylinder like shape and form a bilayered lamellae when exposed to aqueous medium. By addition of cone shape or inverted cone shape species, the properties of the bilayered lamellae can be altered to make it more rigid/leaky. Reprinted from Publication Trends in Biotechnology, 16/7, Dan D Lasic, Novel applications of liposomes, 307-321, Copyright (2020), with permission from Elsevier [16].

Table 1. Classification of different liposome formations.

Types of liposome formations	Commonly used abbreviations	Particle size	Number of lamellae
Small unilamellar vesicles	SUVs	20-100 nm	1
Large unilamellar vesicles	LUVs	>100 nm	1
Giant unilamellar vesicles	GUVs	>1000 nm	1
Multilamellar vesicles	MLVs	>500 nm <sup>a</sup>	>5
Oligolamellar vesicles	OLVs	100-1000 nm	2-5
Multivesicular liposomes	MVVs	>1000 nm	1 (Vesicle inside a vesicle)

<sup>a</sup> This is a typical particle size, however, MLVs with a particle size of 100 nm have been reported [36] J.A. Kulkarni, M.M. Darjuan, J.E. Mercer, S. Chen, R. van der Meel, J.L. Thewalt, Y.Y.C. Tam, P.R. Cullis, On the Formation and Morphology of Lipid Nanoparticles Containing Ionizable Cationic Lipids and siRNA, ACS Nano, 12 (2018) 4787-4795.

### 2.1.1. Solvent evaporation

In this technique, also known as lipid film hydration, phospholipids are dissolved in an organic solvent (more often an equimolar mixture of chloroform and methanol, others could be ether, ethanol, or dichloromethane) [37, 38]. The drug, if lipophilic, is also added to the organic solvent to form a one-phase solution. The organic solvent is subsequently removed slowly under vacuum to form thin sheets of lipid films in which the drug is uniformly dispersed. The thin sheets of lipids are hydrated with an aqueous buffer phase above the glass transition phase of the lipid. If the drug is hydrophilic, it should be dissolved in the aqueous buffer solution. The resulting dispersion gives MLVs with particle size in the micrometer range. This technique is more suitable for lipophilic drugs as a high entrapment efficiency (>90%) can be obtained. For hydrophilic drugs, depending on the physicochemical properties, the entrapment efficiency would routinely around 10-30 % by this passive process. Low entrapment efficiency values have been reported for cytarabine, streptomycin sulphate, chloramphenicol, oxytetracycline, and sulfamerazine [39, 40]. The entrapment efficiency can be increased further for hydrophilic drugs by use of active loading technique. Active loading technique involves transmembrane gradient (like pH or ionic) of unionized species to effect higher entrapment of the drug. A classic example of the active loading technique being used is in manufacture of doxorubicin liposomes. Liposomes are manufactured either by solvent evaporation, solvent dispersion (**Section 2.1.2**) or reverse phase evaporation technique (**Section 2.1.3**) using ammonium sulphate solution as the aqueous phase. The MLVs are further manipulated for size (**Section 2.1.4**) using suitable technique followed by removal of un-entrapped ammonium sulphate using dialysis/diafiltration. The pH of external phase is adjusted (using dialysis/diafiltration) to 7.4 to create a transmembrane pH gradient and a solution of doxorubicin hydrochloride is added that results in entrapment of high amounts of doxorubicin in the aqueous core as precipitates of doxorubicin sulphate [41-43]. Hydrophilic drugs that have been developed into liposomes with high entrapment efficiency using the active loading technique are bupivacaine [44], kanamycin [45], ciprofloxacin [46], chloroquine diphosphate [47], primaquine [48], topotecan [49, 50], and vincristine [51]. For a more detailed theoretical basis readers can refer an excellent book chapter by Boris Čeh [52] and for a more comprehensive preparation techniques for liposomes a review by Has and Sunthar is recommended [53].



### 2.1.2. Solvent dispersion

In this technique the phospholipids are dissolved in an organic solvent that is often miscible with water, ethanol being the preferred solvent [20, 54, 55]. A lipophilic drug would be dissolved in the ethanolic solution together with the phospholipids (other water miscible solvents could be used if the lipophilic drug is not soluble in ethanol). The ethanolic phospholipid/drug solution is added to an aqueous buffer solution, which leads to dilution of the ethanol into the water and thereby spontaneous formation of MLVs. The particle size MLVs is in the micrometer range. This technique is most suited for lipophilic drugs, which can yield high entrapment efficiency. For hydrophilic drugs the entrapment efficiency is normally in the sub 20%. However, as described in **Section 2.1.1**, the entrapment efficiency can be increased significantly reaching >90%.

### 2.1.3. Reverse phase evaporation

This technique is the most preferred techniques for loading a hydrophilic drug in liposomes. For hydrophilic drugs, the internal aqueous core is the only region where the drug can be loaded. Hence, a technique, that can entrap a large amount of aqueous core during formation of liposome will yield a high entrapment efficiency and hence a high drug load. In the reverse phase evaporation method, a w/o emulsion is prepared by dissolving the hydrophilic drug in water and dissolving the phospholipid in water-immiscible solvent (usually chloroform). The organic solvent is then slowly removed, under vacuum, and a gel phase is formed. Further evaporation of the organic solvent yields liposomal dispersion with high entrapment of the aqueous core in the internal core of liposomes. This technique can yield up to 30-50% entrapment of a hydrophilic drug passively [56, 57] and can be increased to >90% using the active loading technique described in **Section 2.1.1**. This technique is discussed extensively by Szoka and Papahadjopoulos [35]. This method and is suited for making small volume parenterals. However, its use on an industrial scale is limited due to the complex manufacturing process.

#### 2.1.4. Size manipulation

Both, solvent evaporation and solvent dispersion, produce MLVs which are in the micrometer range. For drug delivery application, it is important to further reduce the particle size of these liposomes in the submicron range, more specifically in the 50-200 nm, as the particle size of liposomes has a huge impact on the pharmacokinetic and pharmacodynamic profile *in vivo* and hence can have an impact on the therapeutic efficacy of the final formulation [58-60]. On the laboratory scale, there are numerous techniques available that can reduce the particle size of liposomes viz. sonication [8], freeze thaw [61], homogenization [62, 63], and extrusion [64, 65]. All the techniques have their merits and demerits. Sonication is a rather fast technique of reducing the particle size with a high amount of energy dissipated in a small volume. During sonication there is generation of heat, which may lead to degradation of phospholipids and heat labile drugs. Freeze thaw can also be used to convert the MLVs into smaller vesicles as SUVs or LUVs, however, in many cases the it can only reduce the particle size to a certain extent with a rather high particle size distribution, i.e. polydispersity index. Homogenization can also be used for particle size reduction and is a batch process. Liposomes are soft matters that can be reduced in particle size by application of a high-pressure homogenizer. The liposomes obtained by high pressure homogenizer have a higher polydispersity index ( $\sim 0.2$ ) compared to liposomes prepared using extrusion [66, 67] and at high lipid loads ( $>100$  mg/mL) the size reduction efficiency may also be reduced [63]. Extrusion of liposomes through a polycarbonate membrane can produce liposomes of a defined pore size and acceptable PDI ( $\leq 0.1$ ). The extrusion process is a laborious and time-consuming process; however, it is the most acceptable and reproducible process for making liposomes with known defined characteristics.

#### 2.1.5. Final liposomal drug product

The final dosage form, for parenteral administration, of the liposomal drug product can vary depending on the change in physical and chemical properties of liposomes over its storage. It is important to bring the conversation around stability, earlier in the development, as it is leaner to develop all the target product profile early during the development, rather than doing it in late phase development. The factors that one should consider while determining

the final dosage form for liposomes apart from physical and chemical stability is the preferred storage for the commercial product. It is a given that, room temperature storage is preferred over a cold storage for logistical and economic reasons. More often than not the liposomal dispersion stability at room temperature is a challenge and hence most of the marketed products that are presented as a liquid dosage form are required to be stored between 2-8 °C. The particle size, drug loading, and chemical stability are the three important liposomal critical to quality attributes that can get affected during storage of liposomal dispersion as a liquid at room temperature. Another way, to improve the chemical and/or physical stability, at room temperature, is by converting the liposomal dispersion into a dried powder/freeze dried cake for reconstitution. Because of the reduced interaction of the drug product components in the solid state as compared to the liquid state, there is a general trend towards a better chemical and/or physical stability. However, it might not be always possible to have dry liposomal powder for reconstitution due to various reasons. One reason being that often the liposomes do not have the same particle size before and after reconstitution and for these reasons freeze drying/spray drying of liposomes into powders/cake might not be always possible. The second reason being that if the drug is not lipophilic, then it would be inside the liposomal aqueous core and the process of spray drying/freeze drying can change the entrapment efficiency of hydrophilic drugs and the amount of liposome entrapped drug will change after reconstitution. This phenomenon also holds true for drugs which have been loaded in the interior aqueous core of liposomes using active loading technique as in case of doxorubicin loaded liposomes. Hence, in such scenarios, a liposomal dispersion as a liquid that is stored between 2-8 °C is the preferred dosage form. Even, with a better chemical and physical stability profile of a freeze dried/spray dried liposomal product at room temperature, it might still be commercialized as a liquid dosage form to be stored at 2-8 °C. The reason being, that, converting a liposomal dispersion from a liquid dosage form to a solid freeze-dried cake or spray dried powder requires an additional step (along with other in process quality control tests) in the liposome manufacturing under aseptic conditions. Hence, this decision is more driven by the product needs, target product profile, target climatic zones of the marketed product, and the organizational preference of having a cold chain storage or a freeze drying/spray drying capability.

## 2.2. Industrial manufacturing and scale up considerations

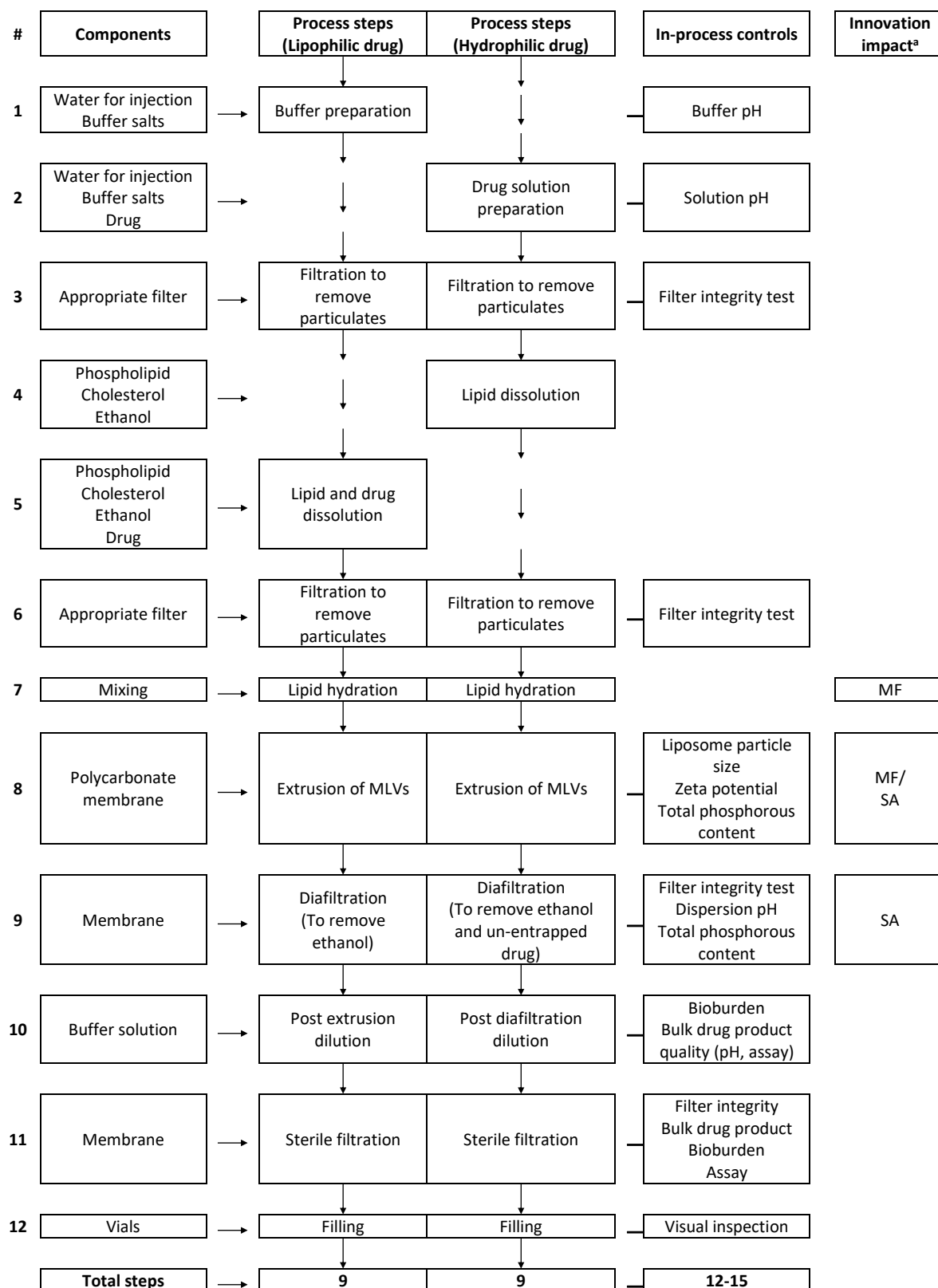
Given that the majority of liposomal formulations are designed to improve drug delivery and reduce off-target toxicity associated with the incorporated cytotoxic drug, the manufacturing process employed must control liposomes' critical quality attributes. These includes particle size (generally < 100 nm), high drug loading and retention (which can be achieved by including high transition temperature lipids and cholesterol) and a near neutral surface charge and/or PEGylation [68].

Despite several methods available for producing liposomes at laboratory scale, there are only a few methods that are used for commercial manufacture that can deliver liposomes with the required critical quality attributes. Of all the methods previously described, ethanol injection followed by extrusion is the most commonly used method of manufacture of large-scale parenteral liposomes. The reason is the reproducibility of liposome particle size and polydispersity index compared to other small-scale manufacturing techniques and the preference of using ethanol (Solvent diffusion) over chloroform (Solvent evaporation) [69]. The particle size and the associated polydispersity index has an influence on the biodistribution and pharmacokinetics of liposomes and hence an impact on the efficacy of liposomes [23]. Hence, a strict control on the particle size is needed, and this is why the extrusion process is critical.

Large-scale manufacturing of liposomes is a long and laborious process and the number of unit operations and associated tests are quite exhaustive (**Fig. 3**). A typical process would involve; buffer preparation, filtration, phospholipid solution preparation, filtration, lipid hydration, extrusion, diafiltration, dilution, sterile filtration, and finally filling. The associated in-process controls for every step increases the complexity of the overall process. A typical quality control would involve pH control at critical steps, filter integrity test, particle size and zeta potential measurements, phospholipid content, bioburden testing bulk drug product assay/pH/related substances, and visual inspection at critical steps. This is a basic large-scale liposome manufacturing process that considers no other additional complexity like active loading as in case of doxorubicin liposomes [70] or freeze drying at the end of manufacturing [71, 72]. Every such step will add to this already complex manufacturing process. If one looks at a typical large-scale manufacturing process of making liposomes (**Fig. 3**) using the ethanol injection method followed by extrusion process, for a model lipophilic or

1 hydrophilic drug, it involves approximately 9-unit operations. And further, the complexity is  
2 increased as every unit operation requires an in-process quality control which makes it a long  
3 and labour-intensive process.

4         Apart from the complex large-scale manufacturing of liposomes, there are numerous  
5 critical to quality attributes of a liposomal drug product that requires careful consideration as  
6 it can affect the end drug product during manufacturing, storage, or its clinical performance  
7 **(Table 2).**



1 Fig. 3. A typical large-scale liposomal manufacturing process.

<sup>a</sup> MF is microfluidics technology and SA is self-assembled vesicular drug delivery systems. Innovation in liposome manufacturing is expected to impact these traditional liposomal manufacturing steps leading to, hopefully, a lean way of liposome production.

1 Table 2. Some critical to quality attributes of a liposomal drug product [1, 73-76].

Drug Product Quality Attributes	Rationale
Appearance	The appearance of liposomal product is affected by free drug and liposomal entrapped drug. The particle size and charge can also affect the appearance of the drug product. Regulatory expectations require appearance testing as a release criterion.
Identification of drug	Identification of the drug is required to ensure patient safety. Administering a drug product with a wrong drug may result in harm to the patient. Regulatory expectations require identification testing as a release criterion. Identification of drug is not affected by any process parameter.
Identification Cholesterol (if added)	Cholesterol prevents leakage of liposomes <i>in vivo</i> as it imparts rigidity of the liposomal membrane. Hence it is critical for the performance of the liposomal drug product <i>in vivo</i> . Identification of cholesterol is not affected by any process parameter.
Identification	Phospholipid is the basic ingredient of liposomes. Identification of phospholipid is not affected by any process parameter.
Assay of drug	In the treatment, adequate plasma levels of drug are essential for an effective engagement of the pharmacological target. In case a lower dose would be administered, the potential for lack of efficacy exists and the impact for the patient would be dependent on the time the patient is not getting the medication. In case of a higher dose, patient safety might be impacted. Regulatory expectations require assay testing as a release criterion.
Drug encapsulation	In some instances, free drug (drug that is not entrapped in liposomes) can show adverse/toxic effects <i>in vivo</i> and in such scenarios, it is important to define and test the drug encapsulation of the liposomal system.
Mean particle diameter	Mean particle diameter impacts the circulation properties of the liposomes. These have consequently an impact on patient safety.
Content uniformity	Content uniformity may impact the dose that the patient is receiving and therefore efficacy of the drug product.
Average pH	pH also impacts the stability of the lipids and hence the release of the drug from the liposomes. Deviating pH values may also cause irritation at the injection site.
Extractable volume	The volume filled in a vial should be such that a complete unit dose can be withdrawn in the clinical setting.
Average osmolality	Deviating osmolality may cause irritation at the injection site. Changes in average osmolality are considered an unlikely cause for irritation to occur.
<i>In Vitro</i> Release	To ascertain the <i>in vitro</i> release performance of the drug product and confirm a lack of uncontrolled leakages under a range of physiological conditions. This should be within the acceptable criteria.
Specific turbidity	Turbidity is controlled by concentration of the lipids and the size of the liposomes. It is linked with the particle size, however, might be more effective in capturing the impact of the bulk properties (including the presence of larger particles) compared to dynamic light scattering (particle size).
Total phosphorous	Total phosphorous is the sum of all the phospholipids. This impacts the liposome properties and drug encapsulation. Free drug, in some instances, can cause severe side effects.
Residual solvents	Residual solvents may cause leakage of drug out of the liposomes. Free drug can cause severe side effects.
Rabbit pyrogen dose	Pyrogens, if parenterally delivered, may lead to significant adverse events including death or multisystem organ failure.
Sterility	Parenteral drug application. Microbial contamination may lead to serious infections

### 2.3. Potential for innovation in large scale liposome manufacturing

Liposomes were first reported in 1960s, and the first liposomal approved product in the market was approved in 1990. Since, then in US and Europe approximately a range of products been approved which use vesicular system as the basis for drug delivery. It seems that, despite the numerous advantages, academic research, and innumerable peer-reviewed publications on liposomes, the total number of liposome/vesicular based products in the highly regulated pharmaceutical space are scarce. Despite the many advantages of liposomes as a versatile drug delivery system, one of the major roadblocks for their commercialization is the difficulty to have a simpler method of making liposomes at laboratory scale as well as commercial scale (as discussed in **Section 2.2**). The current preferred method of liposome manufacture is the ethanol injection method followed by extrusion of the preformed vesicles. The total number of unit operations using the present method of liposome manufacturing is a barrier to a robust formulation and process development. The complexity is immense and hence scale up and/or technology transfer becomes a challenge. Over the years, thousands of research publications have reported different methods of making liposomes which includes, but are not limited to, lipid film hydration, ethanol injection and detergent removal, and a technique that can reduce the overall complexity of liposomes is something that has kept the formulation scientists busy. A lean way of making liposomes will make this drug delivery technology a more attractive prospect in development of new chemical entities as the advantages it offers can make a druggable candidate more druggable and potent, and eventually benefit the patients with reduced total drug load and associated side effects.

As shown in **Fig. 3**, there are numerous steps that are executed during liposome manufacturing. From, an industrial viewpoint, lipid hydration; membrane extrusion; and diafiltration steps are very energy and time intensive as it requires a lot of expertise and allied in process controls. Any technology that can address this problem, either by elimination or modification of the extrusion process can make this a more lean and robust process. The nanoprecipitation/antisolvent technique (of which LeciPlex® is an example), in which the phospholipids and a stabilizer dissolved in a bio-compatible solvent spontaneously forms sub-micron vesicular system when exposed to aqueous environment. These are self-assembled technologies that give a specific particle size and PDI for a given lipid and stabilizer composition. This technology if optimized further could eliminate the extrusion and/or the



1    diafiltration step as the need for solvent removal is not needed and avoids altogether the use  
2    of organic solvents. A different technology that can make liposomes by controlled  
3    precipitation (as in microfluidics) with desired particle size. Microfluidics can eliminate the  
4    need for lipid hydration and extrusion as the vesicles are formed and hydrated in the  
5    microfluidic chamber itself eliminating the need for a separate extrusion step.

6

### 3. Regulatory overview of liposomes

The academical research in liposomes has been very extensive and continues to represent an exciting field of science. Crommelin *et al.* [77] recently made an overview of commercial liposomal drug product marketed in US in the EMA region and reported that 19 products based upon liposome was approved for pharmaceutical purposes, of which 2 have been discontinued, i.e. there is currently 17 marketed products in US and EMA region. Liposomes is used mostly within oncology, but also for fungal infections, pain management, and as carrier systems in vaccine products. Safety is one of the main drivers for the use of liposomes in these cases [77].

Bangham *et al.* first described liposomes in 1961 [78], while the first pharmaceutical product with a liposome-based formulation was approved in 1990, i.e. 30 years later. Doxil, being the first stealth liposome, was approved in 1995, and of the 4 approved stealth liposomes three contains doxorubicin, i.e. conventional liposomes seem to be the preferred formulation strategy for liposomes. Six of the 19 products are freeze dried, hence most is presented as an aqueous dispersion, while stability do not seem to be a general issue. Vyxeos was the first and only liposomal drug product to contain two active compounds, an approach that could be very interesting also for other compounds with synergistic effects.

Of the 19 products that are or have been approved by FDA and/or EMA four contained doxorubicin and two bupivacaine and 3 was three had vaccines, see **Table 3**, while in total 12 small molecules (~70%) are included in a liposome drug product. Doxorubicin liposome formulations is the only generic version approved by FDA/EMA. Worldwide there is a higher amount of generic liposomal products containing doxorubicin or amphotericin B outside the FDA/EMA regions, see **Table 4**. In total 10 and 5 generic liposomal product have been registered for doxorubicin and amphotericin B, respectively. Generic versions of liposomes is still a debated topic in the scientific literature [79, 80], despite the many approvals.

The molecules included in the liposomes had a mean molecular weight on  $559 \pm 271$  g/mol spanning from 243 to 1278 g/mol, so overall aligned with the molecular range for orally administered compounds, though with a shift towards the higher molecular weight range often seen for orally administered compounds [81]. The average log P of the compounds is 1.3 spanning from -3.2 to 5.5, the average melting point is 215 °C ranging from 108 to 255 °C, i.e. compounds formulated in liposomes spans very widely on these two parameters. Hence,

- 1 the versatility of liposomes for drug molecules with varied physicochemical properties is also
- 2 validated by commercially approved products.

1 Table 3. List of commercial liposomal drug products approved in US and/or EMA region. Table modified based upon data from [77].

Product name	Active ingredient	Indication	year <sup>a</sup>	Lipids in formulation <sup>b</sup>	Liposome type	PSD	Finished product	MW (g/mol) <sup>c</sup>	Log P <sup>c</sup>	MP (°C) <sup>c</sup>	AS (mg/L) <sup>c</sup>
AmBisome	Amphotericin	Fungal infections	1990	HSPC:DSPG, chol 2: 0.8 : 1 M	Conventional	< 100 nm	Freeze dried	924.08	0.8	170	750
Doxil/ Caelyx	Doxorubicin	Breast neoplasms; multiple myeloma; ovarian neoplasms; Kaposi's sarcoma	1995	HSPC:chol:DSPE-PEG 56:39:5 M	Stealth	100 nm	Aqueous dispersion	543.52	1.3	229-231	Soluble
DaunoXome	Daunorubicin	Cancer advanced HIV-associated Kaposi's sarcoma	1996	DSPC: chol 2 : 1 M	Conventional	40–80 nm	Aqueous dispersion	527.52	1.8	208-209	39.2
DepoCyt	Cytarabine	Neoplastic meningitis	1999 <sup>d</sup>	DOPC:DPPG	Conventional	20 µm	Aqueous dispersion	243.22	-2.8	186-188	Freely soluble
Epaxal	Inactivated hepatitis A virus	Hepatitis A	1999	DOPC:DOPE 75:25M	Conventional	150 nm	Aqueous dispersion	na	na	na	na
Myocet	Doxorubicin	Breast neoplasms	2000	EPC:chol 55:45 M	Conventional	80-90 nm	Freeze dried	543.52	1.3	229-231	Soluble
Visudyne	Verteporfin	Sub foveal choroidal neovascularization	2000	EPG:DMPC 3:5 M	Conventional	18-104 nm	Freeze dried	718.79	2.1	No data	0.013
DepoDur	Morphine	Pain relief	2004 <sup>e</sup>	DOPC:DPPG	Conventional	17-23 µm	Aqueous dispersion	285.34	0.9	255	149

<sup>a</sup> Year of first approval

<sup>b</sup> Abbreviations used in table: chol: cholesterol; EPC: egg phosphatidylcholine; EPG: egg phosphatidylglycerol; DEPC: 1,2-dierucoylphosphatidylcholine; DOPC: dioleoylphosphatidylcholine; DOPE: dioleoyl-sn-glycero-phosphoethanolamine; DOPS: dioleoylphosphatidylserine; DPPC: dipalmitoyl phosphatidylcholine; DPPG: dipalmitoylphosphatidylcholine; DMPC: dimyristoylphosphatidylcholine; DSPC: distearoylphosphatidylcholine; DSPE-PEG: distearoylphosphatidylcholine polyethylene glycol; DSPG: distearoylphosphatidylglycerol; HSPC: hydrogenated soy bean phosphatidylcholine; SPH: sphingomyelin; PSD: Particle size distribution; MW: Molecular weight; MP: Melting point; AS: Aqueous solubility.

<sup>c</sup> Data from <https://www.drugbank.ca/drugs/>

<sup>d</sup> Product discontinued 2017

<sup>e</sup> Product discontinued 2014

Marqibo	Vincristine	Philadelphia chromosome-negative acute lymphoblastic leukemia	2009	SPH:chol 6:4 M	Conventional	100 nm	Freeze dried	824.97	2.8	220	3 g/L
Mepact	Mifamurtide	Osteosarcoma	2009	DOPC:DOPS 3:7 M	Conventional	1-5 µm	Freeze dried	1277.52	5.5	No data	0.0013
Exparel	Bupivacaine	Anesthetic	2011	DEPC:DPPG:chol:tric apylin	Conventional	24-31 µm	Aqueous dispersion	288.43	3.1	107-108	2400
Lipodox	Doxorubicin	Breast neoplasms	2013	HSPC:chol:DSPE-PEG 56:39:5 M	Stealth	100 nm	Aqueous dispersion	543.52	1.3	229-231	Soluble
Onivyde	Irinotecan	Metastatic Pancreatic Cancer SHARE	2015	DSPC:chol:DSPE-PEG 3:2:0.015	Stealth	110 nm	Aqueous dispersion	586.68	3.2	222-223	Soluble
Mosquirix	RTS, S antigen-based vaccine	Vaccination to help against malaria caused by the parasite Plasmodium falciparum	2015	DOPC:chol	Conventional	50-100 nm	Aqueous dispersion	na	na	na	na
Doxorubicin	Doxorubicin	Breast neoplasms	2017	HSPC:chol:DSPE-PEG	Stealth	100 nm	Aqueous dispersion	543.52	1.3	229-231	Soluble
Nocita	Bupivacaine	Anesthetic	2017	DEPC:DPPG:chol:tric apylin	Conventional	25-31 µm	Aqueous dispersion	288.43	3.1	107-108	2400
Vyxeos	Daunorubicin Cytarabine	Acute myeloid leukemia	2017	DSPC:DSPG:chol 7:2:1	Conventional	107 nm	Freeze dried	527.52 243.22	1.8 -2.8	208-209 186-188	39.2 Freely soluble
Shingrix	Glycoprotein E based vaccine	vaccine for prevention of herpes zoster	2017	DOPC:chol	Conventional	50-100 nm	Aqueous dispersion	na	na	na	na
Arikayce	Amikacin	Mycobacterium avium complex lung disease	2018	DPPC:chol	Conventional	300 nm	Aqueous dispersion	585.60	-3.2	214	50000

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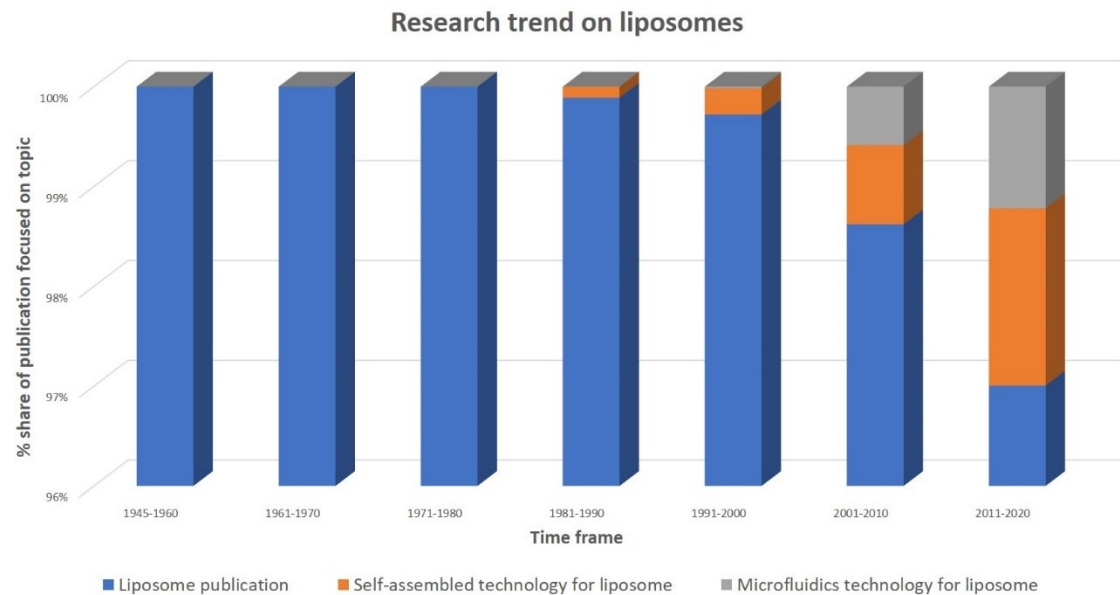
1 Table 4. Generic doxorubicin and amphotericin B liposomal products and their manufacturer.

Drug	Original product	Generic version	Manufacturer
<b>Doxorubicin hydrochloride</b>	<b>Doxil®</b>	Adropeg 20®	Axiommax Oncology Pvt. Ltd
		DOXOrubicin®	Dr. Reddy's
		Doxulip®	United Biotech
		i-dox®	Getwell
		Lipodox®	Sun Pharma
		Lippod™	Celon Labs
		Natdox-LP®	Natco Pharma Ltd.
		Pegadria 50®	Intas pharmaceutical Ltd.
		Rubilong™	Zuventus Healthcare Ltd.
		SinaDoxosome®	Exir Nano Sina Co
<b>Amphotericin B</b>	<b>AmBisome®</b>	Abhope®	Abbott
		Ambilip®	United Biotech
		Amflight™	Celon Labs
		Amphonex®	Bharat serums and vaccines Ltd.
		Phosome 10®	Cipla

2

4. Advances in scalable technologies for liposome fabrication

As discussed in **section 2.2 and section 2.3**, the current manufacturing of liposomes on a large-scale is a challenge as it involves a multi-step multi-test process and that the innovation in this space is much needed. Before, a technology gets explored at the large/commercial scale, it is important to see what has been done in the academic or basic research space to understand the potential solutions and potential pitfalls before larger scale considerations is initiated. A search of self-assembled technology for liposomes and microfluidics for liposomes on PubMed was conducted to understand the trend in the interest of the academia and basic research around liposomes. As seen in **Fig. 4**, basic liposomes research still forms the core of liposomes research (up to 96% publications are focused on basic/traditional liposomes). However, if one looks at the remaining research on liposomes, there has been a steady and increased interest in the non-traditional liposome manufacturing research, and in the last two decades there has been a significant increase in this space. The published literature has indeed focused on alternative ways of making liposomes. In the next sections, we discuss about the liposomes manufacturing technologies that have the potential to simplify the large-scale manufacturing of liposomes and how we can leverage from the literature.



*Fig. 4. Academic and basic research interest in lean liposome manufacturing technique over the years (Ordinate has been adjusted to reflect the microtrend).*

#### 4.1. Self-assembled vesicular drug delivery system

There have been several advancements in the manufacturing techniques of liposomes since their advent in therapeutics. Conventional methods of liposome fabrication involve hydration of a thin film of phospholipids (**Section 2.1.1. and Table 2**) with buffer/aqueous phase to render self-assembly to form vesicular structures [78]. It is necessary that the hydration occurs above the phase transition temperature above which lipids exist in fluid state for self-assembly to take place. Other parameters include concentration of lipids, nature of lipids, volume, type and ionic strength of buffer, temperature of hydration and curing, and agitation time. This section deals with some recently reported novel approaches to render self-assembly yielding vesicular structures.

##### 4.1.1. Heating methods

Nkanga and Krause *et al.* have recently reported liposomes encapsulating cyclodextrin complexes of isoniazid conjugated phthalocyanin prepared by solvent free, easy to scale up heating method [82]. The method involves the use of ethylene glycol, propylene glycol, or glycerol as hydrating adjuvant. Phospholipid and the cyclodextrin-drug complex were hydrated for 60 mins at room temperature with water followed by addition of adjuvant and further stirring for an hour at 70°C during which phospholipids self-assembled to form liposomes encapsulating the complex. Authors reported 58-70% entrapment efficiency of the cyclodextrin-drug complex in liposomes with a particle size of 150 – 650 nm. Entrapment efficiency was observed to be independent of the hydrating adjuvant used. Authors, on the contrary observed a greater entrapment (71%) when liposomes were prepared without the hydrating adjuvant. Liposomes prepared by heating method without hydrating adjuvant exhibited higher entrapment efficiency and also greater size as compared to liposomes fabricated by film hydration method. Surfactant vesicles encapsulating alpha tocopherol have also been recently reported by Basiri *et al.* employing a modification of the heating method proposed by Mozaffari *et al.* [83, 84]. The procedure (**Fig. 5**) involved hydration of surfactants by an aqueous phase at room temperature for 1 h followed by its addition to a preheated (60°C, 5 min) mixture of tocopherol and glycerol. The mixture was further heated at 60°C with stirring (approx.1000 rpm) for a period of 45–60 min under nitrogen atmosphere followed by



sonication as a size reduction step [83]. Authors prepared niosomes with different ratios of Span 60, Tween 60, cholesterol, and dicetyl phosphate. They observed an increase in size with decreasing hydrophilicity of lipid mixture. Around 80% tocopherol could be encapsulated in the niosomes using higher ratios of Span 60: Tween 60 that increased the hydrophobicity of the system. Incorporation of cholesterol to impart rigidity and dicetyl phosphate as charge imparting agent was also thought to contribute to high entrapment. The method utilizes principles of green chemistry, is amenable to scale up, and can be applied to encapsulate both hydrophilic and lipophilic drugs.

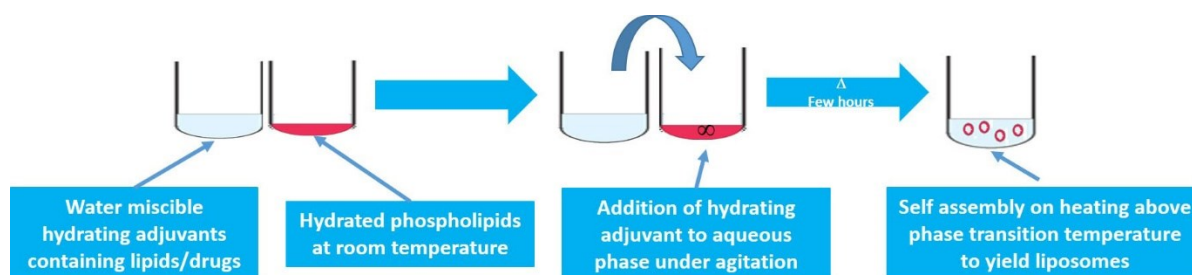


Fig. 5. Schematic representation of heating process to fabricate liposomes.

#### 4.1.2. Nanoprecipitation and Ionic interaction

Recently, Nagarsenker *et al.* fabricated a vesicular system termed LeciPlex® (**Fig. 6**) on the basis of charged interaction leading to enhanced thermodynamic stability using a single step fabrication procedure amenable to scale up. The procedure, a form of solvent dispersion method/anti-solvent method, (**Fig. 7**) involved dissolution of phospholipids in a biocompatible solvent like transcutool HP, that also included a charge imparting agent. The biocompatible solvent phase was heated above the phase transition temperature of the lipids followed by hydration with an aqueous medium at the same temperature. The resultant dispersion contained self-assembled vesicular structures (**Fig. 7**) in the nano range with excellent shelf stability [85]. The system has successfully been explored for encapsulating various drugs like azelaic acid [86], carvedilol [87], idebenone [86], nelfinavir mesylate [88], quercetin [89, 90], silibinin [91], and spironolactone [92]. Authors have also explored various biocompatible solvents such as transcutool, ethanol, glycofurol [93]. Further, depending on the hydrophilicity of the cationic/charge imparting agent, the drug was solubilized either in aqueous phase or organic phase. Authors have reported the system to be very versatile since it has been explored to encapsulate small molecule drugs with molecular weight ranging from

300 Da to 600 Da and with log P value from -2 to 4. Authors have also reported studies with encapsulation of genetic materials like DNA [85].

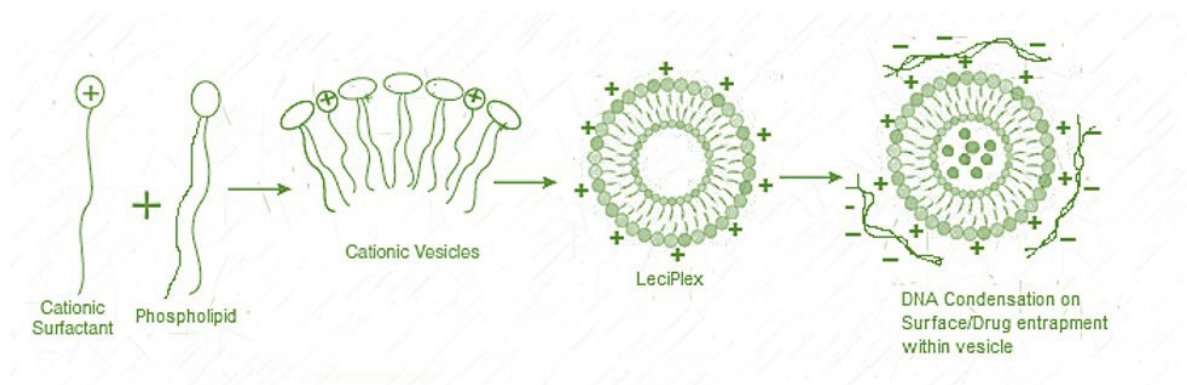


Fig. 6. Schematic representation of LeciPlex® system depicting its possible applications in drug delivery systems.

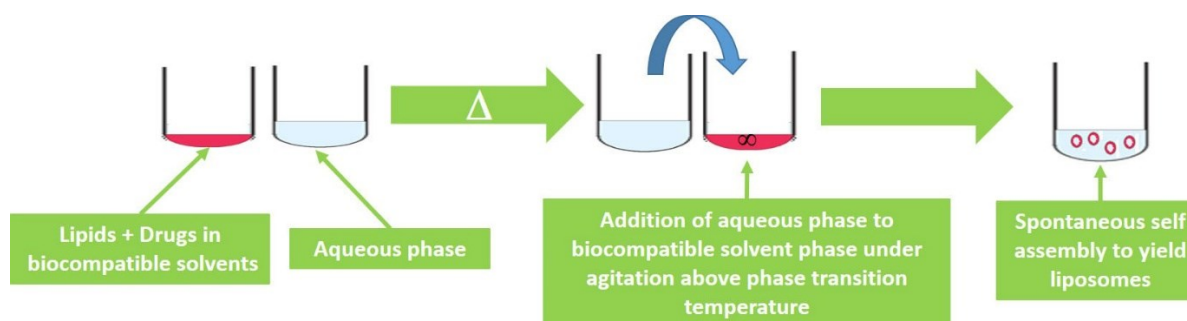


Fig. 7. LeciPlex® fabrication using a single step scalable procedure.

#### 4.1.3. Solvent exchange

Buboltz *et al.* have devised a novel apparatus to fabricate liposomes based on rapid solvent exchange [94]. The apparatus was made up of a tube containing buffer mounted on a vortexer which on actuation formed the buffer into a cylindrical shell. A solution of lipids in an organic solvent was injected under vacuum into an aqueous buffer so that vaporization of the solvent began, along with some evaporative cooling. The vortexing buffer served as a heat reservoir, transferring heat to the droplets to allow vaporization to proceed to completion [94]. Rieder *et al.* have further optimized the protocol for DPPC MLVs by equipping the apparatus with additional controls in regulating temperature, as well as pumping speed, and vortex velocity [95]. Authors conclude that the mechanical forces during vortex mixing and evacuation speeds have greatest effect on formation of liposomes. Rapid evaporation with high vortex speeds resulted in formation of ULVs rather than MLVs. Authors inferred that rapid

evaporation resulted in rupture of lipid membranes, thereby, leading to formation of ULVs. Furthermore, they also observed reducing the vortex speed from 2500 rpm to 600 rpm resulted in formation of concave meniscus at the bottom instead of thin films on the surface. This resulted in foaming which further reduced the evaporation rate leading to formation of MLVs. The overall time for sample preparation was reported to be ~4 mins. Authors also observed that reducing the amount of lipids and the ratio of organic solvent to water resulted in formation of ULVs over MLVs. These factors were more significant contributors when samples were prepared in buffers than water. Authors have explained the formation of ULVs over MLVs on basis of microscopic instabilities present in aqueous phase during liposome formation. These turbulences were thought to provide nucleation points for formation of lipid vesicles where self-assembly takes place. Greater microturbulences resulted in more nucleation points and hence amount of lipid present per unit point reduced resulting in formation of ULVs over MLVs.

#### 4.1.4. High Shear

Recently, Anderson *et al.* have patented a protocol to prepare liposomes using high shear method [96, 97]. Briefly, the process involved dispersing dried powder of lipids in a suitable buffer. Using an equipment that provides high shear, the dispersion was heated to a temperature above the phase transition temperature. Initially a low shear was applied to avoid foaming. After the phase transition temperature was reached, mixture was stirred at high shear until a desired size distribution was reached which was followed by cooling to room temperature. Authors reported an average size of liposomes to be  $163.2 \pm 0.493$  nm with a PDI of 0.258. The method has been patented for fabrication of immunogenic liposomes containing vaccine adjuvants [96, 97]. Shen *et al.* have also studied the effect of high shear to a surfactant solution containing MLVs to produce ULVs. They observed the surfactant system without shearing to contain ULVs and MLVs with a size range of 300 – 500 nm. Homogenization of the system at 200 bar resulted in conversion of MLVs to ULVs of the size range of 50 – 75 nm. Authors, however, observed an increase in size of this ULVs over 9 days of storage. Cryo-TEM images revealed the ULVs to be potato shaped rather than spherical in nature. Authors attributed this observation to emulsification at high pressure which caused ULVs to be in a unstable state leading to increase in size after storage in a bid to reach low energy state [98].

Wang *et al.* have reported glass beads to produce shear to reduce size of liposomes in nanoscale as opposed to conventional methods, where glass beads were only used to increase the surface area for film formation. The method comprised of dissolving lipids in chloroform followed by solvent evaporation by rotary evaporation to yield a thin film to which aqueous solution of drug was added along with glass beads followed by prolonged shaking to yield vesicles [99, 100]. Wang *et.al* developed liposomes in nano range (60 to 550 nm) by employing glass beads of different sizes to yield shear. The size of liposomes was observed to increase with the increase in diameter of glass beads used during preparation. The smaller glass beads (2 mm) possessed less density and therefore did not provide adequate shear forces resulting in small fraction of large vesicles (800 nm) along with small sized vesicles (100 nm). Increasing the time of shear to 24 h, however resulted in further lowering of size (67 nm) with good PDI. Further, use of large glass beads (5 mm) lead to formation of liposomes of greater size with good PDI. With increased time of shear, a reduction in size of liposomes to 100 nm was observed. Authors reported 3 mm and 4 mm beads to produce liposomes of 100 nm size with 1 h of shearing, however PDI values remained high. The authors also observed 2 mm and 5 mm glass beads to yield best entrapment efficiencies of amphotericin B (Up to 92%) [99].

#### 4.1.5. Emulsification and solvent evaporation

Suzuki *et al.* have recently reported a multiple emulsification-solvent evaporation method to prepare liposomes that can yield higher entrapment efficiency for hydrophilic moieties [101]. The process (**Fig. 8**) comprised of primary emulsification to formulate a water in-oil (w/o) emulsion that contained the drug and a volatile organic solvent containing a mixture of bilayer-forming lipids. This was followed by a secondary emulsification step to obtain w/o/w emulsions effected by means of microchannel emulsification technique. The primary emulsion was forced through channels in an aqueous phase to form the w/o/w emulsion followed by evaporation of solvent to yield self-assembled lipid vesicles entrapping the hydrophilic drug moieties. Authors reported the method to possess a wide control over particle size range of vesicles with 0.2  $\mu\text{m}$  to several micron size particles being obtained. The size of the liposomes obtained was observed to be dependent upon the globule sizes of primary emulsion and the technique used to effect emulsification. The mean diameters of water droplets in the primary w/o emulsions was 0.2  $\mu\text{m}$  with probe sonication, 1.2  $\mu\text{m}$  with

an ultrasonic bath sonication, and 4.4  $\mu\text{m}$  via extrusion through PTFE membrane. The vesicle size remained unchanged during the microchannel emulsification step and in the final drug product. The type of surfactant did not affect the size of vesicles. Authors therefore concluded size of the primary emulsion droplets to be a determinant of final vesicle diameters. Entrapment efficiency was observed to be a function of droplet size and type of surfactant. A larger w/o emulsion droplet size lead to a reduction in entrapment of drug. [101]. Kuroiwa *et al.* have employed similar technique to produce MLVs wherein the primary emulsion was obtained by sonication followed by multichannel emulsification to yield the double emulsion [102, 103]. Kuroiwa *et al.* have also observed size of vesicles to be directly related to size of primary w/o emulsion droplets. To investigate the importance of secondary emulsification, vesicles without second emulsification step were fabricated. The final dispersion contained particles larger than 1  $\mu\text{m}$  size with entrapment efficiencies as low as 55%. Using microchannel emulsification and sodium caseinate as emulsifier up to 82% entrapment of calcein was obtained. A high entrapment efficiency was observed with sodium caseinate as compared to Tween 80. The authors attributed these high entrapment values to microchannel emulsification technique which offers advantages of forming multiple emulsion droplets under low shear and without heating reducing the leakage of hydrophilic materials from internal aqueous compartment to external aqueous phase [103].

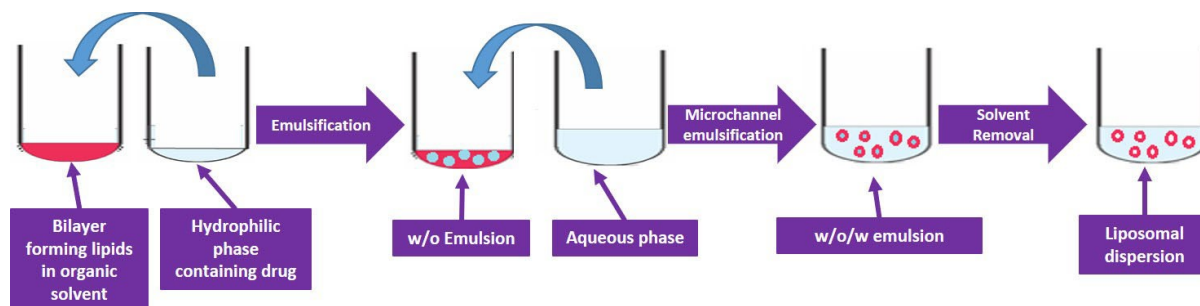


Fig. 8. Schematic representation of emulsification solvent evaporation method for entrapment of hydrophilic drugs.

#### 4.1.6. Packed bed-based reactors

Liu *et al.* have designed a rotating packed bed reactor for continuous manufacturing of liposomes [104]. The apparatus consisted of a rotor, a mesh packing, two liquid inlets, and a liquid outlet. High centrifugal force was created on the packing by the rotor. Solvent phase and aqueous phase were pumped into the reactor through the liquid inlets and sprayed on the inner edge of the rotor by using several holes in the liquid distributor at the centre of the

reactor. Liposomes were manufactured by injecting methanolic solution containing lipids and aqueous phase into separate inlets of the reactor at increasing temperatures maintained by circulating water in the jacket of the tanks. Liposome suspension was collected from the reactor outlet and then dialyzed against PBS (pH at 7.4) to remove residual organic solvent. Authors observed flow rate ratio (FRR) of solvent phase to aqueous phase to be one of major factors determining particle size. The flow rate of organic solvent was maintained at 20 mL/min and aqueous phase flow rate was increased from 20 mL/min to 300 mL/min. An increase in the FRR intensified the two-phase velocity difference, thereby, enhancing mass transfer and nucleation leading to formation of small size liposomes. The high gravity level (HGL), a parameter, calculated by the authors as a measure of centrifugal force generated by the rotor was another critical process parameter affecting liposome size. Increasing HGL also resulted in enhanced micromixing and greater mass transfer as the fluids were split into thin streams and tinier droplets. The temperature did not seem to affect the size of the liposomes. In contrast, a slight increase in the particle size was observed from 208 nm to 232 nm when the temperature was increased from 20 °C to 50 °C. A further increase in temperature to 60 °C did not affect the particle size. The authors attributed this particle size increase to fusion of liposomes at higher temperature. An increase in entrapment efficiency of sorafenib was observed on increasing the temperature from 20 °C to 40 °C after which a decrease in entrapment efficiency was observed when temperature reached 60 °C. Authors attributed the reduction in entrapment efficiency to hydrolysis of phosphatidylcholine at higher temperature thereby disrupting lipid bilayers and causing leakage of drug. Authors also report a similar trend of FRR on entrapment efficiency. A lower FRR resulted in reduced contact between the two phases with drug being retained in organic phase due to higher solubility leading to low entrapment efficiency. An increase of FRR above an optimum value resulted in decreased contact between lipids and drug resulting in low entrapment efficiency. Using optimal conditions, authors could obtain liposomes with mean particle size of 200 nm and entrapment efficiency of 89%. The authors also report an output of 33.6 kg/day of drug loaded liposomes under optimum conditions thereby suggesting this technique to demonstrate high potential for liposome production in large scale [104].

#### 4.1.7. Gel assisted self-assembly

Weinberger *et al.* have devised a novel method of liposome formation in solid state assisted using PVA [105]. Authors prepared a 5% (w/w) PVA solution and coated microscope coverslips by spreading 100–300  $\mu\text{L}$  of PVA solution onto it followed by oven drying. Lipids solubilized in chloroform were spread on the dried PVA film and placed under vacuum for 30 min for evaporation of the solvent. Buffer was placed in a Vitrex chamber formed on cover slip and GUV formation was tracked using phase contrast microscopy. Authors have utilized the fabrication method to encapsulate proteins in liposomes [105].

#### 4.1.8. Spray drying and fluid bed drying

Conventional spray drying has been used by many researchers to prepare liposomal dispersions. Maniyar *et al.* prepared liposomes using a onestep spray drying process as previously reported in the literature [106]. Briefly, lipids were dissolved in methanol: chloroform (1:1) solvent system to which drug was added and finally lactose was added as a carrier to yield the dispersion for spray drying. The dispersion was then subjected to spray-drying with the inlet and outlet temperatures set to 80 °C and 50 °C respectively and feed rate of 5 mL/min. The spray flow rate was set to 1.5 kg/cm<sup>2</sup>. Authors reported the liposomes to be of 270 nm size with a PDI of 0.239 and an entrapment efficiency of 56.38% [106].

Gala *et al.* have reported a novel approach utilizing methods that are industrially feasible such as fluid bed coating, high pressure homogenization, and freeze-drying [107]. Briefly, authors prepared pro-liposomes by spraying the ethanolic solution of phosphatidylcholine onto sucrose particles in a fluid bed coater. Authors performed the coating and drying at temperature as low as 30 °C to avoid lipid melting and thereby particle agglomeration and to ensure proper spreading of wet phospholipid on the sucrose carrier particles. The process was completed in two hours and yields was reported as high as 20% w/w from the original weight of sucrose. The pro-liposomes were further hydrated for two hours at 60 °C and freeze dried to yield the final product. Sucrose is thought to be advantageous in its dual role which it plays as a carrier in the formulation of proliposomes and as a cryoprotectant during freeze-drying. Pro-liposomes and liposomes generated after hydration were nanosize which were further size reduced to the range of 70 – 125 nm using high pressure homogenization and freeze dried. Authors report that the freeze-drying of the nano-liposomes retained the size below 155 nm post reconstitution. An increase in

entrapment efficiency of beclomethasone dipropionate was also observed, probably, due to an increased interaction between drug and lipids on removal of water. Presence of residual ethanol was stated to cause interdigitations in liposome bilayers resulting in very poor entrapment efficiencies and hence authors also suggest the drying time to be at least 2 hours to evaporate all the ethanol [107].

Nirale and Nagarsenker have also explored the possibility of preparation of liposomes by spray drying a methanolic solution containing phospholipids and lactose dissolved in it. Spray dried powder on hydration with saline yielded giant vesicles of size ranging from 800 nm to 6  $\mu$ m while hydration with saline yielded liposomes of mean size of 3  $\mu$ m [108].

#### 4.1.9. Solvent diffusion-based methods

Many researchers have attempted to advance the traditional solvent based method to render it a single step process to yield the final product. In this regard, Costa *et al.* have reported a modification of the conventional ethanol injection method [109]. The equipment consisted of three pressurized tanks containing lipid solutions in ethanol, which were pumped under a controlled rate to a static mixer, which ensured proper mixing of all lipids prior to it reaching the injection port where the organic and aqueous streams converged. Authors reported flow rate of 5 to 40 mL/min for organic phase and 60 to 400 mL/min for the aqueous phase. The entire process was automated using computer algorithms where the user has to define final lipid concentration and molar ratios of lipids. Liposomes that were formed were unilamellar, monodispersed and possessed a size of ~25 nm to >465 nm depending on the lipid type and flow rate [109]. Pulseless flow rates, Reynolds number of mixed ethanol/aqueous flow stream and FVR were three parameters that determined the formation of jet and governed the PDI of liposomes. Low FVR and low Reynolds number resulted in a stratified stream and limited mixing leading to formation of polydisperse liposomes. High Reynolds number along with low FVR lead to formation of weak jet thereby yielding polydisperse liposomes. Maintaining a high FVR results in monodisperse liposomes, with size being governed by Reynolds number. Further, the size was also observed to be more dependent on flow rate of aqueous phase rather than lipid concentration.

Another novel inline method that integrates all processes involved in liposome preparation has been developed by Araki *et al.* [110]. The equipment consists of the in-line



1 thermal mixing device with modified counterflow dialysis to yield in-line closed liposome  
2 production system. The process comprised of dissolving lipids with aid of heat and drugs in  
3 isopropanol followed by dilution with maltose and a sodium phosphate solution. This  
4 dispersion was delivered to the in-line thermal mixing device. The solution during heating was  
5 passed through 0.22  $\mu\text{m}$  filter thereby achieving sterilization followed by cooling which  
6 resulted in self-assembly of lipids to form liposomes. The heating and cooling temperatures  
7 were set at 80 °C and 20 °C respectively. The dispersion was then subjected to a counterflow  
8 dialysis against buffer solution to remove the organic solvent and concentrate the dispersion.  
9 The liposomes were further freeze dried using polysorbate 80 as the cryo-protectant. Authors  
10 obtained a monodisperse liposomal vesicles of 100 nm size using this process without an  
11 additional homogenization step. Decreasing the amount of organic solvent was shown to  
12 reduce the size of the liposomes due to increased hydrophobic interactions between  
13 phospholipid molecules. Authors state that solubility of the lipids determines the amount of  
14 organic solvent to be used which bears an influence on the size of the liposomes. Authors  
15 reported the process to be scalable with a scale-up production that can be set up with a simple  
16 parallel processing. Authors further reported that the process bears an aseptic production  
17 capability which is amenable to complete automation without additional human intervention  
18 [110].

#### 4.1.10. Freeze drying.

21 Recently, Liu *et al.* have prepared liposome using a lyophilization monophasic solution  
22 technique [111]. The technique as reported involves dissolving the lipids, drug, and  
23 lyoprotectants in a TBA/water system followed by freeze-drying to yield the pro-liposomes.  
24 On hydration, these give the liposomal dispersion. This method is a one-step process and is  
25 amenable for large-scale liposome preparation. Using this technique, Liu *et al.* prepared  
26 liposomes encapsulating glycyrrhetic acid. Briefly, authors dissolved drug and lipids in TBA  
27 at 45 °C, and lyoprotectant was dissolved in 45 °C water. When mixed at appropriate ratios,  
28 these two solutions produced a third clear isotropic monophasic solution. This solution was  
29 sterilized by filtration and lyophilized. Prefreezing was performed for 12 h at – 40 °C followed  
30 by primary drying at temperature of – 50 °C for 24 h with a chamber pressure of 1–20 Pa.  
31 Authors report the entrapment efficiency and particle size of the reconstituted liposome to  
32 be 72.82% and 198 nm respectively. The dispersion was stable for 6 months at 25 °C. Authors

observed that entrapment efficiency was affected by amount of phosphatidylcholine, amount of cholesterol, and volume percentage of TBA; while particle size was observed to be more dependent on volume percentage of TBA. Further, sublimation rate during lyophilisation was observed to increase with increasing volume percentage of TBA [111].

#### 4.1.11. Supercritical fluid techniques

Recently, researchers have also employed supercritical fluids to assist in liposome preparation [112]. The process is called as Supercritical assisted Liposome formation (SuperLip) [113]. Lipids were dissolved in ethanol which were then mixed with pure carbon dioxide in a saturator to obtain an expanded fluid. The saturator was filled with baffles, maintained under high pressure and thermally heated by thin bands to produce the supercritical fluid. The mixture was fed to a high-pressure formation vessel to which an aqueous solution in an atomized form containing the drug was also introduced. The working temperature of the saturator and the formation vessel was set to 40 °C and pressure to 100 bar. The liposome suspension was recovered from the bottom of the vessel; a decompression step was used to separate carbon dioxide and ethanol using a stainless-steel separator which was maintained at 30 °C and 10 bar. The technique has been explored to encapsulate hydrophilic drugs like theophylline where up to 98% entrapment efficiency has been reported. Multilamellar liposomes of ~200 nm size with >90% entrapment efficiencies for lipophilic drugs have been reported by this technique. The encapsulation efficiency was observed to depend on flow rate of the aqueous solution with reduced flow rates resulting in higher values of entrapment. Higher flow rates of aqueous phase caused greater velocity of droplets after atomization, and thereby, impacted the disruption of the droplets on the receiving container at the bottom. This resulted in lower entrapment efficiencies of theophylline. Lower flow rates in contrast produced bigger droplets resulting in dispersion with smaller mean size but larger size distribution. Trucillo *et al.* have also investigated this technique to prepare liposomes with three antioxidants, Farnesol, Linalool, and limonene [114]. Lipid soluble antioxidants were loaded in a stabilized emulsion which was atomized and introduced in the formation vessel containing the ternary mixture viz. CO<sub>2</sub>/ethanol/Lipids to obtain the formation of liposomes. Farnesol, linalool, and limonene were the anti-oxidants investigated for encapsulation. Two approaches were used for encapsulating these molecules in liposomes. In the first approach, the lipids along with anti-oxidants were dissolved in ethanol while in second approach, lipid

1 soluble antioxidants were dissolved in isopropyl myristate and loaded in a stabilized emulsion.  
2 The aqueous phase/emulsion phase was atomized and introduced in the formation vessel  
3 containing the ternary mixture viz. CO<sub>2</sub>/ethanol/Lipids to obtain the formation of liposomes.  
4 Authors attribute various parameters such as atomization in a high-pressure environment,  
5 turbulent shear forces, and viscosity of the medium to affect liposome formation and the size.  
6 Authors report liposomes to be smooth, homogenous with a size range of 300 – 600 nm and  
7 entrapment efficiency of >90% prepared using this technique [114]. **Table 5** list array of/  
8 number of all the techniques that involve self-assembly of liposomes.

1 Table 5. Comparison of different methods based on self-assembly with respect to manufacturing and scalability.

Method	Process Parameters to be controlled	Size Range	Drug type	Liposome type	Scalability	Comments
Conventional Thin film hydration	<ul style="list-style-type: none"> <li>Hydration time</li> <li>Hydration temperature</li> <li>Sonication time.</li> </ul>	>1 $\mu\text{m}$	Lipophilic	GUVs	Presently used commercially	<ul style="list-style-type: none"> <li>Multi step process.</li> <li>Homogenization/Extrusion is required post fabrication for size reduction</li> <li>Sterilization operation post fabrication/aseptic processing is required.</li> <li>Can be “tens of” to “hundreds of” litres on large scale</li> </ul>
Conventional Ethanol injection	<ul style="list-style-type: none"> <li>Rate of injection</li> <li>Volume of solvent/aqueous phase</li> <li>Agitation rate</li> </ul>	>100 nm [54]	Lipophilic & hydrophilic [20]	SUVs, MLVs	Presently used commercially	<ul style="list-style-type: none"> <li>Pilot plant scale design using this method has been reported with liposomes [54].</li> <li>Removal of ethanol using rotary evaporator is an added step.</li> <li>Up to 3L batch has been prepared [54].</li> <li>Can be “tens of” to “hundreds of” litres on large scale</li> </ul>
Heating methods	<ul style="list-style-type: none"> <li>Temperature</li> <li>Time of hydration and self-assembly</li> </ul>	>1 $\mu\text{m}$	Lipophilic	GUVs	+++	<ul style="list-style-type: none"> <li>Multiple step process.</li> <li>Size reduction is needed post fabrication to attain desired size.</li> </ul>
Nanoprecipitation	<ul style="list-style-type: none"> <li>Volume of aqueous and organic phase</li> <li>Agitation rate and time</li> <li>Rate of addition</li> <li>Temperature of operation</li> </ul>	200 to 500 nm [85]	Lipophilic	GUVs, OLVs	+++++	<ul style="list-style-type: none"> <li>Simple fabrication procedure.</li> <li>Use of biocompatible solvent is an advantage</li> <li>Size reduction operation may be required subsequently</li> <li>Sterilization operation post fabrication/aseptic processing is required.</li> </ul>
Solvent exchange method	<ul style="list-style-type: none"> <li>Temperature and pressure during operation</li> <li>Pumping and vortexing speed</li> <li>Volume of solvent</li> </ul>	>1 $\mu\text{m}$	Lipophilic	MLVs	+	<ul style="list-style-type: none"> <li>Rapid process</li> <li>Polydispersity of product is high.</li> <li>Size reduction techniques are needed post fabrication to attain desired size.</li> </ul>
High Shear method	<ul style="list-style-type: none"> <li>Temperature</li> <li>Time of heating</li> <li>Shear force (e.g. speed of rotor)</li> </ul>	~200 nm (good PDI) [98]	Charged drugs	SUVs	+++++	<ul style="list-style-type: none"> <li>Simple easy to scale up technique</li> <li>Avoids use of organic solvents.</li> </ul>
Emulsification evaporation	<ul style="list-style-type: none"> <li>Homogenization technique to prepare primary emulsion</li> <li>Microchannel emulsification equipment dimensions</li> </ul>	>1 $\mu\text{m}$	Lipophilic & hydrophilic	MLVs, GUVs [71]	+++	<ul style="list-style-type: none"> <li>Multi step process.</li> <li>Leakage of hydrophilic drug during fabrication may result in less entrapment.</li> <li>Size reduction/sterilization operation is required post fabrication.</li> </ul>

	<ul style="list-style-type: none"> <li>Flow rates of primary emulsion and aqueous phase to yield secondary emulsion</li> </ul>					
Packed Bed Reactors using high gravity technology	<ul style="list-style-type: none"> <li>Flow rate of aqueous and organic phases</li> <li>Temperature</li> <li>Type of packing</li> <li>Rotational speed of the reactor</li> </ul>	~100 nm (good PDI) [115]	Lipophilic	SUVs	+++	<ul style="list-style-type: none"> <li>Reduced number of unit operations</li> <li>Provides liposomes of desired size</li> <li>Removal of residual solvent is necessary before further processing.</li> </ul>
Gel assisted methods	<ul style="list-style-type: none"> <li>Surface of gel formed</li> <li>Temperature set during process</li> </ul>	>1 µm [105]	gene/peptide /monoclonal antibodies?	GUVs	+	<ul style="list-style-type: none"> <li>Easy process yields rapid formation of GUVs without need for special equipment</li> <li>Scale up may be challenging</li> </ul>
Spray drying/Fluid bed coating	<ul style="list-style-type: none"> <li>Inlet and outlet temperature</li> <li>Spray flow rate</li> </ul>	100 nm to >1 µm [106-108]	Lipophilic	MLVs	+++	<ul style="list-style-type: none"> <li>Process yields pro-liposomes;</li> <li>Yields are however low.</li> </ul>
Freeze drying	<ul style="list-style-type: none"> <li>Pre-freezing time and temperature</li> <li>Primary drying conditions</li> <li>Secondary drying conditions</li> </ul>	~200 nm (good PDI) [111]	Lipophilic	SUVs	+++	<ul style="list-style-type: none"> <li>Pre-filtration of solution followed by lyophilisation can result in sterile final product with reduced operations.</li> </ul>
Modified solvent-based methods	<ul style="list-style-type: none"> <li>Feed rates of organic and aqueous phase</li> <li>Temperature during operation.</li> </ul>	50 to 500 nm [109, 110]	Lipophilic	SUVs	+++++	<ul style="list-style-type: none"> <li>Process has a smaller number of unit operations to yield desired liposomal size</li> <li>Ease of sterilization is advantage.</li> <li>Removal of residual solvents is necessary from final product.</li> <li>Up to 4L batches have been attempted [110].</li> </ul>
Supercritical fluid assisted fabrication	<ul style="list-style-type: none"> <li>Temperature and Pressure of formation cell and separator chambers</li> <li>Feed rates of carbon dioxide and aqueous solutions</li> <li>Dimensions of nozzle to produce atomization.</li> </ul>	100 to 200 nm (good PDI) [114]	Lipophilic & hydrophilic	MLVs	+++	<ul style="list-style-type: none"> <li>Process can be used to encapsulate hydrophilic and lipophilic drugs</li> <li>Process has a smaller number of unit operations but is complex as compared to other techniques</li> <li>Sterilization may be achieved by pre-filtration of solutions.</li> </ul>

## 4.2. Microfluidics

The application of microfluidics for the manufacture of liposomes has gained considerable academic interest over recent years, as shown in **Fig. 4**. Microfluidics offers the ability to consistently produce optimised, uniform nanoparticles [116]. The production can be scale-independent [117], allowing translation of formulations from laboratory to GMP [118], addressing many of the limitations of traditional bulk production methods. In liposome production, microfluidics can replace the lipid hydration and extrusion steps during liposome production and replace it with a single-step process for liposome production where particle size is process controlled (**Fig. 3**). The ability to achieve this is related to the process offering the miniaturisation of the fluidic mixing environment; by using intersecting microchannels, nanolitre volumes of fluids are mixed in a highly controlled format [119, 120]. A key feature of many of the microfluidics systems is the generation of laminar flow, which is challenging to produce in macroscale systems. Using laminar flow, nanoprecipitation can be easily controlled. Single phase mixing systems are the most commonly adopted due to process simplicity. These systems control the mixing of two or more miscible solvents (commonly an aqueous phase mixed with a water miscible alcohol such as methanol, ethanol or isopropanol). During the mixing, the change in polarity promotes nanoprecipitation and the formation of liposomes [121] (**Fig. 9**). This process can also be referred to an 'anti-solvent approach' [122]. A hypothesis for self-assembly of liposomes resulting from this process was proposed by Zook and Vreeland [123]. Within their model, as the alcohol and aqueous phase mix, the polarity of the mixture increases. With this increase in polarity, the lipids become progressively less soluble and the lipid monomers self-associate into planar bilayer discs. As these discs increase in size, this increases the surface area of the hydrophobic chains around the edge of the discs that is in contact with the polar solvent. To reduce this hydrophobic surface area, the discs bend and form spherical liposomes [123]. By incorporating water soluble drugs in the aqueous phase and/or lipid soluble drugs in the solvent phase the drug can be simultaneously incorporated into the liposomes. Using this method, the ability to entrap small molecules [124, 125], nucleic acids, [126] and proteins [17] within liposomes have been demonstrated. Further, it has been demonstrated that the method is a scale-independent production of liposomal adjuvants [15, 127, 128].

#### 4.2.1. Microfluidic cartridge design

Fluid flow can occur in two different ways: laminar flow or turbulent flow. The type of flow produced depends on the velocity and viscosity of the fluid and mixing in macroscopic flow is generally turbulent [129]. However, microflows are more commonly laminar, and mixing under standard conditions involves molecular diffusion processes only, which is inefficient [129-131]. Therefore, to address this and enhance mixing efficiency, micromixers have been optimised with regard to channel geometry and architecture such that effective mixing can be achieved within short mixing channels and with high throughput [132]. Indeed, a key factor in the formation of liposomes using microfluidics is the residence time within the mixing chamber as well as the geometry of the microfluidic mixer as this controls the rate of nanoprecipitation and vesicle formation.

A wide range of micromixers have been developed and tested for liposome production, including toroidal mixers, staggered herringbone mixers, T-mixers and hydrodynamic flow focusing (See **Fig. 9** and **Table 6**). For example, hydrodynamic flow focusing can be achieved in 2D or 3D. With the 2D device, fluid is injected concurrently through three inlets. A central stream of a water-miscible solvent containing lipids is focused horizontally by aqueous fluid streams introduced perpendicular to the central stream [133, 134]. With this system, lipid aggregation and liposome formation can occur at the wall of the microfluidic channel [134]. This can present issues in particle size control and micromixer fouling/blocking [133, 135]. To address this, 3D hydrodynamic flow focus mixing has been developed. Within this system, the alcohol solvent stream is introduced by a capillary surrounded by the aqueous stream. This avoids aggregation of the nanoparticle components at the walls of the mixer [134, 136]. Hood *et al.*, demonstrated the use of hydrodynamic flow focusing to produce liposomes and within their microfluidic cartridge they also included microdialysis for buffer exchange and establishment of a pH gradient which supported drug loading of doxorubicin to give complete on-chip production [137].

### Few types of microfluidic mixers

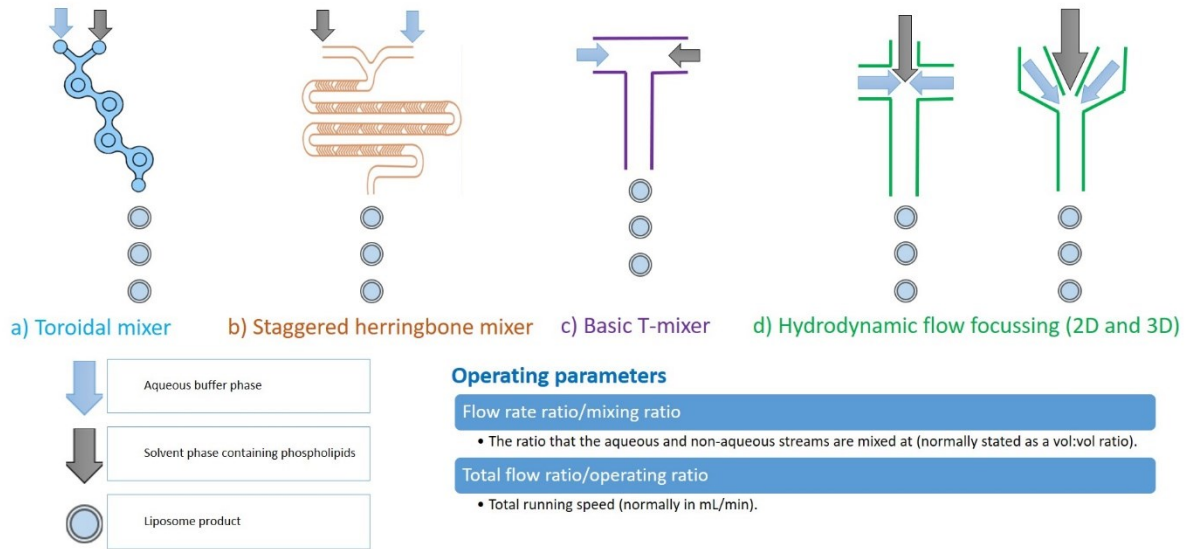


Fig. 9. Schematic representations of example micromixer cartridge designs that can be used. A) a toroidal mixer with planar geometry employing centrifugal forces to encourage uniform mixing, b) the staggered herringbone micromixer with embossed chevrons, c) a basic T-mixer with two inlets where fluids are forced into a T junction and d) hydrodynamic flow focussing with three inlets where a central stream of solvent is focused by aqueous fluid streams either in 2D or 3D.



1 **Table 6.** Examples of microfluidic mixers used to produce liposomes. Table modified based upon data from [117].

Microfluidic architecture	Formulation <sup>a</sup>	Entrapped material	Loading	Particle size	Reference
Toroidal mixer	DSPC:Chol DSPC:Chol:PS DOPE:DOTAP	OVA OVA PolyA	26 - 36 % 15 – 25 % 95 – 100 %	50-60 100-120 40-60	[117]
Staggered herringbone mixer	POPC:cholesterol	Doxorubicin	60 to 100 % depending on formulation	20 – 30	[138]
	DLinkE2-DMA:DSPC:Chol:PEG-DMA	si-RNA	95 – 100 %	30 - 55	[139]
	DSPC:Chol	Metformin and Glipizide	20 – 25 % 38 – 44 %	50 - 60	[124]
	DMPC:Chol / DSPC:Chol	Atenolol and quinine	100% 50 – 80 %	200 – 360	[140]
	Tween85:Chol:DDA	siRNA	Not determined	70 – 230	[141]
	ATX:DSPC:Chol:DMG-PEG:PEG2000	si-RNA	Not determined	40 – 50	[142]
T-mixer	Triolein:POPC:DSPE-PEG2000	Iron oxide	0.43 w/w	35 - 140	[143]
Hydrodynamic flow focussing	DMPC:Chol:DPPE-PEG2000	Doxorubicin	Up to 72%	80 - 190	[137]

2  
3 To increase the efficiency of mixing, the contact area between the aqueous and  
4 solvent phases can be increased through appropriate microchannel configuration [132].  
5 Passive mixers, such as the staggered herringbone mixer, have been developed to achieve  
6 this and have been used to manufacture a range of liposome formulations (**Table 6**). The  
7 staggered herringbone design has an in-groove pattern in the microchannel with an  
8 asymmetric herringbone shape [144, 145]. As a result of this structure, the fluid streams are  
9 passed over a series of protruding herringbone structures causing chaotic flow. This creates  
10 transverse vortices that are repeatedly changed because of the asymmetric geometry [146].  
11 This micromixer construct can be used to produce size-controlled liposomes with their size  
12 being tightly defined by process parameters [138]. However, the complex structure of this  
13 micromixer present practical limitations in terms of fabrication costs, and throughput speeds  
14 [117]. To address this, an alternative design based on a toroidal mixer design has been  
15 developed. This micromixer promotes laminar flow at high fluid speeds by using circular

<sup>a</sup> Abbreviations. DSPC: 1,2-distearoyl-sn-glycero-3-phosphocholine; Chol: Cholesterol; PS: phosphatidylserine; DOPE: 1,2-dioleoyl-sn-glycerol-3-phosphoethanolamine; DOTAP: 1,2-dioleoyl-3-trimethylammonium-propane; POPC: 1-palmitoyl-2-oleoyl-sn-glycero-3-phosphocholine; DLinkE2-DMA and ATX: proprietary lipids; PEG-DMA: N-[(methoxy poly(ethylene glycol)2000)carbonyl]-1,2-dimyristoylpropyl-3-amine; DMPC: 1,2-dimyristoyl-sn-glycero-3-phosphocholine; DMG-PEG2000: 1,2-Dimyristoyl-sn-glycerol, methoxypolyethylene glycol-2000; DSPE-PEG: 1,2-distearoyl-sn-glycero-3-phosphoethanolamine-N-[methoxy-(polyethylene glycol)-2000]; DPPE-PEG2000: 1,2-Dimyristoyl-sn-glycerol methoxypolyethylene glycol -200.

structures within the flow path. This induces chaotic advection through increasing the number of vortices and centrifugal forces created between the columns within the cartridge, allowing for improved mixing and higher throughput [131]. Using this new toroidal micromixer design, the production of liposomes from laboratory-scale to GMP scale was demonstrated [117].

In addition to reduced complexity and time for manufacture, the use of microfluidics offers further advantages. For example, size-tuned production of liposomes (from 40 to 500 nm depending on the formulation and process parameters e.g. [17, 138, 148]) with low polydispersity (often below <0.1 PDI) can be achieved through the control of process parameters. The use of microfluidics has also been shown to promotes higher loading within liposomes compared to other commonly used methods; for example, a forty-fold increase in iron oxide loading was achieved using a T mixer compared to other methods [143]. A five to ten-fold increase in protein loading was also achieved when microfluidics was adopted as the manufacturing process compared to liposomes produced by lipid hydration and sonication or extrusion [17]. This was achieved with small homogenous (50 – 60 nm; <0.2 PDI) liposome formulations that could be scaled from bench scale to GMP production rates [117]. An additional advantage of this microfluidic production method is that it can be conducted at ambient temperature negating the need to work above lipid transition temperatures, as is normally needed for lipid hydration based methods [17]. In the case of complexation of nucleic acids within lipid nanoparticles, whilst loading is generally 100% irrespective of the production method used, microfluidics offers controlled complexation and size tuned production (e.g. [139]). With liposomes where high drug loading is achieved using active loading (e.g. doxorubicin loading), microfluidics offer the opportunity to support liposomal drug formulations in a reduced time, with minimal reagent waste [137].

#### 4.2.2. Microfluidic material and production parameter considerations

During the production process, the focus is on optimising the key driving forces of vesicle assembly including the component solubilities, concentrations and process parameters. As with the other manufacturing methods discussed, the selected process parameters can impact on the end product attributes which includes size, size distribution, lipid composition, and drug loading/release characteristics (**Table 7**).

1     *Table 7. Material and process considerations to consider in microfluidic production of liposomes and LNPs.*

Considerations when developing microfluidic production processes			
Materials considerations		Process parameters	
<b>a) Buffer</b>	Aqueous buffer strength; buffer strength can be used to control particle size [128]	<b>a) Production flow rates</b>	Flow rate can be used to control particle size [140]
<b>b) Solvent selection</b>	<ul style="list-style-type: none"> <li>a. Suitability for large scale production [69].</li> <li>b. Lipid(s) solubility; lipid concentration can also impact on particle size and drug loading [124].</li> <li>c. Solvent polarity; polarity can impact on particle size and drug loading [147].</li> </ul>	<b>b) Aqueous to alcohol mixing ratio</b>	Mixing ratio can be used to control particle size. Mixing ratio can also impact on drug loading and drug release [17, 138, 148, 149]
		<b>c) Operating temperature</b>	<ul style="list-style-type: none"> <li>a. Microfluidic production of liposomes does not need to be conducted above the transition temperature of lipids [17]</li> <li>b. Heat can improve lipid solubility in solvents [150]</li> </ul>

2  
3

#### 4.2.2.1. Aqueous buffer selection

When considering the initial fluid attributes, the choice of aqueous phase, alcohol phase and lipid concentration all impact on the liposome product attributes. For example, recent work [128] has shown that by controlling the aqueous buffer concentration, the particle size of cationic and anionic liposomes could be controlled. Through controlling the buffer concentration of Tris in the aqueous phase, highly monodisperse, cationic liposomes at selected size between 40 nm and 500 nm were produced. *In vivo* biodistribution studies in mice also showed that by using this method to creating small (<50 nm) liposomes, the clearance rates of these liposomes from the injection site was increased and increased accumulation to the draining lymphatics promoted.

#### 4.2.2.2. Alcohol solvent selection

The alcohol selection in the production process is a key factor to consider as the suitability for scale-up manufacturing and the solubility of the selected lipids in the solvent must be considered. Working with solvents that have low toxicity potential and defined as class 3 in the ICH Q3C (R6) [69] (e.g. ethanol and IPA) is preferable followed by those in class 2 (e.g. methanol). However, the choice of solvent can have an impact upon liposome attributes [147]. Results show that in general, reducing the polarity of the solvent (e.g. by replacing methanol with isopropanol) increased the liposome particle size and reduced drug loading. However, the choice of solvent did not impact on liposome short-term stability or drug release characteristics. By using solvent combinations such as methanol/isopropanol mixtures to modify solvent polarity, the resultant liposome particle size was also similarly modified. However, not all liposome formulations were sensitive to the impact of solvent choice; liposomes containing charged lipids and formulations containing increased concentrations of cholesterol or pegylated-lipids were less sensitive to solvent choice [147]. In the proposed model of liposome self-assembly during microfluidics [123], vesicle size is determined by two factors: the growth rate of planar bilayer discs and 2) the rate the discs close into spherical vesicles. As the alcohol and buffer mix, polarity of the mixture increases and lipid discs form. Thus, depending on the polarity of the alcohol adopted, the overall polarity of the mixture running through the micromixer can be reduced. Thus, if the solvent is switched from methanol to isopropanol, this may result in larger lipid discs forming (**Fig. 9**) and subsequently larger liposomes form. In terms of lipid concentration, some studies suggest that low initial

lipid concentrations tend to promote larger and more heterogeneous liposome suspensions and generally, at higher initial lipid concentrations there is no impact (e.g. [17, 127]). However, the relationship between lipid concentration and particle size is not clear and it maybe formulation and microfluidic cartridge design dependent. For example, in studies using single hydrodynamic focusing, increasing lipid concentrations increased particle size [151, 152]. This trend was also shown with the production of LNPs using a chaotic mixer [153]. Within this study [153], that authors note the size of lipid discs formed (and the subsequent size of vesicles) is controlled by the rate of mixing, which controls the ethanol concentration and mixture polarity. With a more rapid reduction in ethanol concentration, smaller vesicles formed [153]. The authors also propose that at high lipid concentrations, larger lipid discs form at the saline-ethanol interface promoting larger LNP formation [153].

#### 4.2.2.3. Process parameters – aqueous to alcohol mixing ratio

Like the impact of solvent selection, the rate of mixing of the aqueous and alcohol during the production of liposomes impacts on liposome attributes. Generally, low mixing ratios tend to form larger vesicles; for example studies by Kastner *et al.* using liposomal formulation DOPE:DOTAP and PC:Chol on a staggered herringbone micromixer chip demonstrated a reduction in average liposome size (from 200 nm to 50 nm) as the alcohol content in the mixture reduced from 50% to 17% [125, 126]. Jahn *et al.* also showed that as the flow rate ratio increased (and the alcohol content in the mixture reduced), the resulting particle size decreased [121]. Zizzari *et al.* also demonstrated this size-control effect for the liposomal formulation HSPC:Chol:DSPE-PEG-2000 produced over a range of flow rate ratios [154]. There are two possible rationales why the polarity gradient produced can be used to control liposome particle size. Jahn *et al.* [155] propose that during microfluidic mixing, the initially formed liposomes may take-up alcohol within the bilayer. This can promote in some partial disassembly of the liposomes. As the solvent phases continue to mix, then the alcohol concentration in the liposomes will decrease, resulting in re-assembly. When the flow rate ratio is such that the overall alcohol concentration is low, this may limit the assembly / re-assembly cycle and reduce liposome size [155]. Alternatively, Zizzari *et al.* note that at higher flow rate ratios, a smaller solvent stream results and as the lipid discs formed at the liquid interface, they may bend and forming liposomes more rapidly in the presence of decreasing solvent concentration. The length of time these lipid discs can grow will directly impact upon

the final resulting particle size, with shorter times leading to smaller liposomes. Thus at high alcohol content flow rate ratios (e.g. 1:1) the longer the time available for lipid discs to expand and for larger vesicles to form [156]. Interestingly the mixing flow rates was also shown to impact on drug loading and drug release properties, with lower alcohol concentrations in the flow mix producing smaller liposomes, with lower protein loading and higher release rates [17].

#### 4.2.2.4. Process parameters – flow rate

The operating flow rate adopted in microfluidics is an important consideration as it will dictate production speeds. The flow rate though the micromixer can have an impact on liposome size; however, this tends to be micromixer and formulation dependent. For example, Sedighi *et al.* [157], use design of experiments to rapidly screen and optimise various liposomes formulations. They tested the impact of both flow rate ratio and total flow rate. From their studies using a staggered herringbone mixer, they reported that the flow rate ratio had a significant impact on particle size and size distribution (as covered in section 4.2.1.3), whilst liposome characteristics remained constant at flow rates above 8 mL/min. We have also recently shown the ability to produce liposomes with mapped characteristics (size, polydispersity and drug loading) with flow rates from 12 to 200 mL/min using a toroidal mixer [117]. However, with cationic liposomes prepared from dimethyldioctadecylammonium bromide (DDAB) and trehalose 6,6'-dibehenate (TDB), their vesicle size was shown to be controlled by both the flow rate ratio and total flow rate [127]. With these systems, increasing the flow rate and flow rate ratio was shown to reduce sizes from around 1000 nm (1:1 flow rate ratio and 5 mL/min flow rate) down to 160 nm (5:1 flow rate ratio, 20 mL/min flow rate). However, due to their cationic nature the different liposome sizes showed similar protein loading and clearance from the injection site after intramuscular injection [127].

#### 4.2.2.5. Process parameters – operating temperature

In many liposome production methods, liposome manufacture is undertaken at temperatures above the transition temperature of the lipid bilayer [158]; for example, DSPC liposomes are commonly prepared above 55°C. This can present issues for thermos-labile drugs, particularly proteins. However, when microfluidics is employed, liposomes can be produced at ambient temperature irrespective of their lipid composition [17, 117]; studies have shown that using

microfluidic production, liposomes produced at room temperatures or at temperatures above the main lipid transition temperature are similar in size with DSPC:Cholesterol liposomes of the same size being produced at operating temperatures from 20 to 60°C. This demonstrates there is no requirement to work above the lipid transition temperature during the microfluidic manufacturing process [17]. This may be a result of alcohol being present within the liposomal membranes during the initial production stage. When interacting with membranes, ethanol and other short chain alcohols can locate in the headgroup region where the hydrogen bonds between the alcohols and the phosphate and carbonyl groups of the lipids can form [159, 160]. This changes the packing in the lipid membrane and can increase membrane fluidity [161]. Fatty alcohols have also been used as an alternative to cholesterol in the formulation of liposomes and can reduce the transition temperature of liposome bilayers in a similar manner to cholesterol [162]. However, it can be useful to use elevated temperatures to improve the solubility of some lipids in solvents during the processes irrespective of their  $T_c$  (e.g. [127]).

#### 4.2.2.6. Scale-independent production, continuous manufacturing using microfluidics and down-stream processing considerations.

Microfluidics offers the advantage of being scale-independent. This allows rapid optimisation of liposomes characteristics to be undertaken using a design-of-experiment approach [156], followed by translation of the formulation and production parameters from laboratory production through to continuous manufacturing (which can be described as a processing concept whereby product constantly flows out). Continuous manufacturing has a long history in the non-pharmaceutical industries and has been adopted for active pharmaceutical ingredients and solid oral dosage forms such as ORKAMBI™, PREZISTA®, VERZENIO™ and SYMDEKO®. The advantage of continuous manufacturing includes, lower capital expenditure, smaller factory footprint and lower Cost of Goods. For lower volume production, semi-continuous can also be adopted. Continuous manufacturing has evolved from bulk drug, to solid dosage forms through to more complex biologicals. Therefore, the transition of this to complex formulations is the next step. The development of down-stream processing can also support this and selected relevant processing techniques and principles can be applied to the production of liposomes. Liposome manufacturing has some inherent aspects that make scalable manufacture and continuous production appropriate. Liposomes have some

elements of solid oral products (drug manufacturing/sourcing/supply chain) and some from biopharma (mixing vessels, tangential flow filtration, filtration etc) and some unique elements (particle size, size distribution and drug loading) [163]. For example, the application of TFF can be applied to liposomal manufacture. TFF offers two key processing steps for liposomes 1) purification (removal of solvent and/or free drug) and 2) concentration adjustment. TFF processes can be run in a recirculating loop or as a single pass format, where multiple TFF cassettes are run in series. This can offer lower system hold-up volumes. After purification of the system, sterilisation can be achieved by incorporating in-line sterile filtration as part of the process train.

As part of a continuous or semi-continuous production process, process analytical technology should be included to analyse and control the manufacturing process and to monitor the liposome critical quality attributes. Microfluidics offers opportunities in terms of design flexibility, process control and parameter predictability. It also offers ample opportunities for modular production setup, process feedback and process control [164]. This requires rapid in-line or at-line methods to monitor product attributes such as particle size analysis and drug loading. However, currently there is no process analytical methods available for microbial contamination, so sterility assurance would need to be assured through the design and validation of the system. For batch production, this can be achieved by testing a single bulk sample prior to filling and capping. If continuous filling is used, then a representative bulk is not applicable. Therefore, further developments are needed in these areas to allow fully continuous production to be adopted.



## 5. Conclusion

Liposomes, and lipid-based nanomedicines play a key role in healthcare. Their ability to protect, deliver and target drugs provides enhanced efficacy and reduced toxicity, which have been explored for both classical small molecules as well as for RNA delivery. However, as a complex drug product their manufacture presents challenges with multi-batch processes commonly used. This limits their wider application and challenge their general applicability as it complicates the development of a robust, scalable and affordable process, that may challenge the value proposition of the product in some diseases and regions. With the development of new manufacturing processes, which offer leaner manufacturing and scale-independent manufacture, the application of liposomes and other lipid-based nanomedicines can be more readily translated from the pre-clinical research through to production and clinical use. To support this, rapid on-line and at-line analytical tools are required that can support the characterisation and quality assurance of the drug product.

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4

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