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Near and vacuum UV polarization spectroscopy of 1,4-distyrylbenzene

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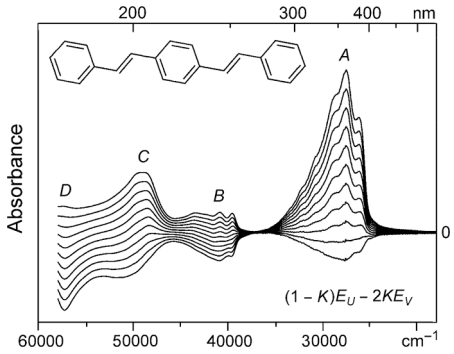
HIGHLIGHTS

- UV absorbance of 1,4-distyrylbenzene (DSB) measured with synchrotron radiation.
- Linear Dichroism (LD) spectroscopy in the near and vacuum UV spectral regions.
- Stretched polyethylene as an anisotropic solvent.
- Semiempirical and Time-Dependent Density Functional Theory (TD-DFT) calculations.

GRAPHICAL ABSTRACT



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ABSTRACT

The UV absorbance bands of 1,4-distyrylbenzene (1,4-Bis[(*E*)-2-phenylethenyl]benzene, DSB) are investigated by Synchrotron Radiation Linear Dichroism (SRLD) spectroscopy using stretched polyethylene as an anisotropic solvent. The observed polarization data provide information on the transition moment directions of the observed spectral features. The investigation covers the range 15,000–58,000 cm^{-1} (667–172 nm), thereby providing new information on the transitions of DSB in the vacuum UV region. The observed spectrum is characterized by four main band systems centered at 27,600, 41,000, 49,800, and 57,500 cm^{-1} (362, 244, 201, and 174 nm). In general, the observed bands and their polarization directions are well predicted by the results of quantum chemical calculations using Time-Dependent Density Functional Theory (TD–DFT) with the functional CAM-B3LYP, and with the semiempirical all-valence-electrons method LCOAO.

1. Introduction

The spectroscopic and photophysical properties of 1,4-distyrylbenzene (1,4-Bis[(*E*)-2-phenylethenyl]benzene, DSB, Scheme 1) and its derivatives have been the subject of several investigations [1–18]. DSB is "an important prototype medium-sized π-conjugated organic compound for optoelectronics" [17] and has been characterized as "one of the working horses of optical spectroscopy" [14].

We have previously investigated the ground state absorption (GSA) spectrum of the closely related chromophore 1,4-bis(phenylethynyl)

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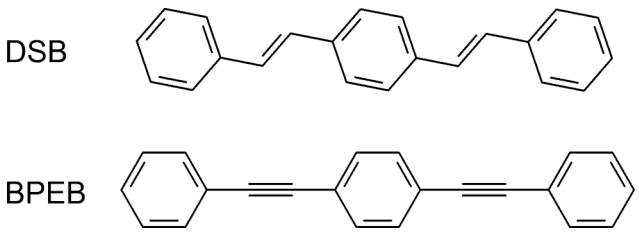


Scheme 1. 1,4-distyrylbenzene (DSB) and 1,4-bis(phenylethynyl)benzene (BPEB).

benzene (BPEB, Scheme 1) [19]. In the present publication, we report the results of a similar study of DSB. The absorbance spectrum is investigated by UV Synchrotron Radiation Linear Dichroism (SRLD) spectroscopy on partially aligned molecular samples using stretched polyethylene (PE) as an anisotropic solvent. LD polarization spectroscopy on oriented molecular assemblies yields information on the polarization directions of the recorded absorption bands [20–25], information that frequently leads to resolution of otherwise overlapping spectral features. The use of synchrotron radiation [19,26,27] allows extension of the measurements up to about 58,000 cm^{-1} (172 nm), thereby providing evidence for previously unobserved transitions of DSB in the vacuum UV region. The observed spectral features are compared with theoretical electronic transitions computed with the semiempirical all-valence-electrons models ZINDO [28,29] and LCOAO [30,31] and with Time-Dependent Density Functional Theory (TD-DFT) [32–35] using the functionals LC-ωPBE [36,37], ωB97XD [38], and CAM-B3LYP [39]. For previous theoretical calculations of the electronic transitions of DSB, see literature referenced by Roldao et al. [17]. Additional information is provided as Supplementary data, referred to in the ensuing text as S1–S6.

2. Experimental

2.1. Sample preparation

1,4-distyrylbenzene (DSB) [CAS 1608–41-9] (>95 %) was purchased from ChemDiv Inc. The spectroscopic purity of the substance was checked by comparison with literature spectra [4,6,17,40,41]. Low-density polyethylene (PE) was obtained as pure 100 μm sheet material from Hinnum Plast, Denmark. DSB was introduced into the PE by submersion of a piece of the polymer sheet into a saturated solution of the compound in chloroform (Merck Uvasol) at 50 °C for a week. To accelerate the process, the solution was sonicated in an ultrasound bath for thirty minutes each day. Subsequently, the chloroform was allowed to evaporate from the doped sample and crystalline deposits on the surface were removed with methanol (Merck Uvasol). The sheet sample was finally uniaxially stretched by ca. 500 %. A reference sample without solute was produced in the same manner. Further details on stretched PE samples can be found in the literature [20–25].

2.2. Linear Dichroism (LD) spectroscopy

Two linearly independent absorbance curves were measured at room temperature, one with the electric vector of the sample beam parallel to the uniaxial stretching direction (*U*), and one with the electric vector perpendicular to it (*V*); in both cases the sample beam was perpendicular to the surface of the PE sheet. The resulting baseline-corrected LD absorbance curves $E_U(\tilde{\nu})$ and $E_V(\tilde{\nu})$ are shown in Fig. 1 (top).

The 15,000–47,000 cm^{-1} (667–213 nm) region was recorded using a Shimadzu UV-2101PC spectrophotometer equipped with rotatable Glan-Taylor prism polarizers in sample and reference beams. The instrument was validated by an internal calibration procedure. The 37,000–58,000 cm^{-1} (270–172 nm) region was measured with synchrotron radiation as previously described [19] on the CD1 beamline [26,27] at the storage ring ASTRID at the Centre for Storage Ring Facilities (ISA). The same physical sample was used in both experiments; combination of the two data sources was unproblematic. Validation was undertaken by checking the overlap region of the two recorded spectra.

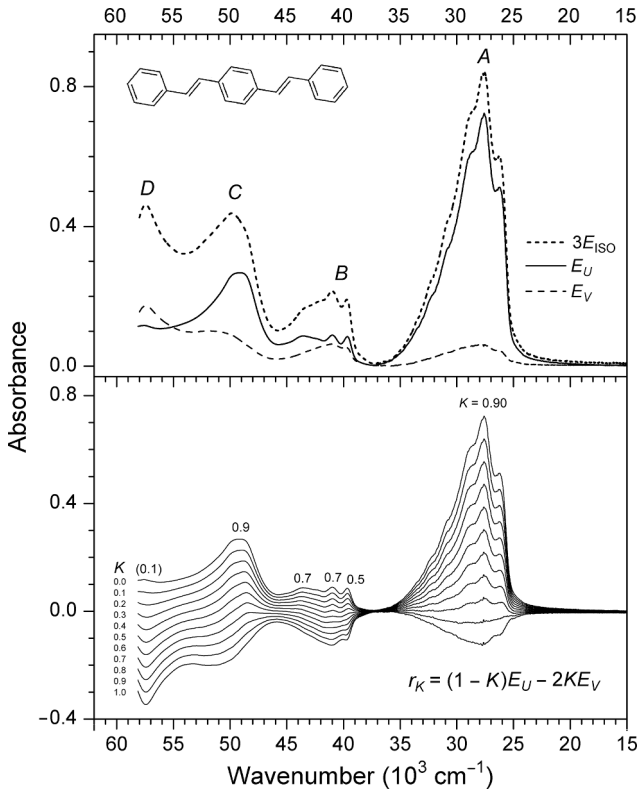


Fig. 1. *Top:* Linear Dichroism (LD) absorbance curves $E_U(\tilde{\nu})$ and $E_V(\tilde{\nu})$ for 1,4-distyrylbenzene (DSB) in stretched polyethylene measured at room temperature with the electric vector of the sample beam parallel and perpendicular to the uniaxial stretching direction *U*. $3E_{ISO}(\tilde{\nu}) = E_U(\tilde{\nu}) + 2E_V(\tilde{\nu})$ is three times the absorbance that would have been measured in an isotropic experiment on the same sample. *Bottom:* Family of reduced absorbance curves $r_K(\tilde{\nu}) = (1 - K)E_U(\tilde{\nu}) - 2KE_V(\tilde{\nu})$ with *K* varying from 0 to 1 in steps of 0.1.

3. Calculational details

Quantum chemical calculations of the electronic transitions of DSB were carried out with the semiempirical all-valence-electrons models ZINDO [28,29] and LCOAO [30,31], and with TD-DFT procedures adopting the long range-corrected density functionals LC-ωPBE [36,37], ωB97XD [38], and CAM-B3LYP [39]. The ZINDO and the TD-DFT calculations were performed with the Gaussian 16 software package [42], the LCOAO calculation with the computer program published in Ref. [43].

The TD-DFT calculations considered vertical transitions from the electronic ground state to the 110 lowest excited singlet states, using the basis set AUG-cc-pVTZ [44,45]. The ground state molecular equilibrium geometries were optimized with the respective functionals under the assumption of C_{2h} symmetry. In the optimizations performed with LC-ωPBE and CAM-B3LYP, dispersion effects were represented by the model by Grimme [46], empiricaldispersion = gd3bj [42] (the ωB97XD functional includes empirical dispersion [38]). In all the DFT and TD-DFT calculations, the influence of the solvent medium was approximated by the Polarizable Continuum Model IEFPCM [47–50] using solvent = *n*-hexadecane [42]. The C_{2h} equilibrium coordinates predicted with CAM-B3LYP/AUG-cc-pVTZ are given in S1. One very small imaginary torsional frequency equal to *i*2.4 cm^{-1} is predicted (S1), indicating that the assumption of planarity is acceptable. A selection of electronic

transitions predicted with TD–CAM-B3LYP is listed in Table 1, complete results are provided in S2.

The calculations performed with the ZINDO method considered excitations to the 110 lowest excited singlet states [42]. The LCOAO calculation included interaction between all singlet configurations generated by promotion of an electron from the 11 highest occupied to the 11 lowest unoccupied π-type molecular orbitals (MOs). The LCOAO calculation provided prediction of MCD B-terms [20,31,51] for the computed transitions. The C_{2h} molecular input geometry for the ZINDO and LCOAO calculations was taken as the one predicted with B3LYP [52,53] using the basis set cc-pVTZ [44,45]. The LCOAO results are provided as S3 and S4 (a complete LCOAO bibliography is included as S5).

A graphical comparison of the electronic transitions predicted with ZINDO, LCOAO, TD–LC-ωPBE, TD–ωB97XD, and TD–CAM-B3LYP is shown in Fig. 2. The convolutions of the predicted transitions were performed by assigning a Gaussian function to each excitation wavenumber with an area proportional to the oscillator strength of that transition, using a constant standard deviation, σ = 1500 cm^{-1}. The molecular orbital (MO) surface diagrams shown in Fig. 3 were computed with CAM-B3LYP/AUG-cc-pVTZ and visualized using GaussView 6 [54] with a constant isovalue equal to 0.02.

4. Results and discussion

4.1. Linear dichroism: orientation factors and polarization directions

The observed LD absorption curves $E_U(\tilde{\nu})$ and $E_V(\tilde{\nu})$ are shown in Fig. 1 (top). The investigated spectral range is characterized by four main band systems: *A*, *B*, *C*, and *D* centered at 27,600, 41,000, 49,800, and around 57,500 cm^{-1} (362, 244, 201, and 174 nm) (Table 1); the energy position of the latter band is quoted with less accuracy, due to its close proximity to the spectrophotometers low-wavelength cut-off. The directional information that can be obtained from the LD curves is represented by the orientation factors K_i for the transition dipole moments of the observed transitions [20–24]:

$$K_i = \langle \cos^2(\boldsymbol{M}_i, U) \rangle \quad (1)$$

Here $(\boldsymbol{M}_i, U)$ is the angle of the moment vector $\boldsymbol{M}_i$ of transition *i* with the stretching direction *U* of the polymer. The pointed brackets indicate the average over all solute molecules in the light path. A large K_i indicates that the transition moment is efficiently aligned with the stretching direction, and vice versa. The K_i values may be determined by the graphical TEM (Trial and Error Method) procedure [20–22]. We consider the reduced absorbance curves $r_K(\tilde{\nu})$ [22]:

$$r_K(\tilde{\nu}) = (1-K)E_U(\tilde{\nu}) - 2KE_V(\tilde{\nu}) \quad (2)$$

The contribution from transition *i* vanishes from the linear combination $r_K(\tilde{\nu})$ for $K = K_i$. Provided a spectral feature due to transition *i* can be identified in both $E_U(\tilde{\nu})$ and $E_V(\tilde{\nu})$, the K_i value may thus be determined by visual inspection [22]. A family of curves $r_K(\tilde{\nu})$ for DSB with *K* ranging between the limits 0 and 1 is shown in Fig. 1 (bottom). The estimated orientation factors *K* for the observed spectral features are listed in Table 1.

A well-defined *K* value equal to 0.90 is determined for the strong band *A* peaking at 27,600 cm^{-1} (362 nm). The large value indicates that the moment vector is well aligned with the stretching direction. An even larger *K* equal to 0.95 was determined for the corresponding transition in the spectrum of BPEB [19].

Band *B* displays three closely spaced peaks at 39,700, 41,000, and 43700 cm^{-1} (252, 244, and 229 nm) with smaller *K* values close to 0.5, 0.7, and ~0.7. The value for the latter peak is approximate because of the broadness of this feature. A close-up of the family of reduced absorbance curves for band *B* is provided in S6.

The bands *C* and *D* in the vacuum UV region with maxima near 49,800 and 57,500 cm^{-1} (201 and 174 nm) are relatively broad and

Table 1
Observed spectral features and calculated electronic transitions for 1,4-distyrylbenzene (DSB).

Experimental					TD–CAM-B3LYP/AUG-cc-pVTZ				
	$\tilde{\nu}$[a]	$3E_{ISO}$[b]	*K*[c]	$\lvert\phi\rvert$[d]	Term[e]	$\tilde{\nu}$[a]	*f*[f]	ϕ[d]	Leading configurations[g]
A	27.6	0.85	0.90	(0°)	1^1B_u	27.8	**1.95**	(0°)	93 % [$6b_g \rightarrow 6a_u$], 5 % [$5a_u \rightarrow 7b_g$]
					2^1A_g	36.0	0	–	49 % [$6b_g \rightarrow 7b_g$], 44 % [$5a_u \rightarrow 6a_u$]
					2^1B_u	37.4	0.06	+5°	40 % [$6b_g \rightarrow 7a_u$], 33 % [$4b_g \rightarrow 6a_u$]
					3^1B_u	39.6	0.01	+23°	23 % [$6b_g \rightarrow 8a_u$], 21 % [$5b_g \rightarrow 6a_u$]
					3^1A_g	39.8	0	–	28 % [$6b_g \rightarrow 8b_g$], 23 % [$4a_u \rightarrow 6a_u$]
					4^1A_g	41.3	0	–	43 % [$6b_g \rightarrow 7b_g$], 41 % [$5a_u \rightarrow 6a_u$]
					1^1B_g	41.8	0	–	68 % [$6b_g \rightarrow 33a_g$], 16 % [$6b_g \rightarrow 34a_g$]
B	39.7	0.19	0.5	43°	4^1B_u	43.4	**0.30**	+24°	40 % [$4b_g \rightarrow 6a_u$], 14 % [$6b_g \rightarrow 7a_u$]
	41.0	0.22	0.7	29°	5^1B_u	44.4	**0.28**	+61°	19 % [$5b_g \rightarrow 6a_u$], 22 % [$6b_g \rightarrow 9a_u$]
	43.7	0.17	0.7	29°	6^1B_u	46.3	**0.18**	+45°	29 % [$6b_g \rightarrow 9a_u$], 25 % [$5b_g \rightarrow 6a_u$]
					7^1B_u	47.5	0.01	+27°	50 % [$3b_g \rightarrow 6a_u$], 23 % [$6b_g \rightarrow 9a_u$]
C	49.8	0.44	(0.9)	(0°)	8^1B_u	50.9	**0.24**	−8°	15 % [$5a_u \rightarrow 7b_g$], 15 % [$6b_g \rightarrow 10a_u$]
					9^1B_u	51.8	**0.87**	−7°	19 % [$4b_g \rightarrow 7a_u$], 13 % [$6b_g \rightarrow 10a_u$]
					10^1B_u	52.9	0.02	−5°	33 % [$5a_u \rightarrow 7b_g$], 22 % [$6b_g \rightarrow 10a_u$]
					11^1B_u	54.4	0.01	−71°	18 % [$5b_g \rightarrow 6a_u$], 12 % [$5a_u \rightarrow 8b_g$]
					12^1B_u	54.6	0.06	−29°	19 % [$6b_g \rightarrow 10a_u$], 17 % [$4b_g \rightarrow 7a_u$]
					13^1B_u	55.1	0.01	−79°	58 % [$6b_g \rightarrow 11a_u$], 12 % [$5a_u \rightarrow 9b_g$]
D	(57.5)	0.47	(0.1)	(76°)	14^1B_u	57.0	**0.40**	+58°	29 % [$5a_u \rightarrow 8b_g$], 21 % [$4a_u \rightarrow 7b_g$]
					15^1B_u	58.0	**0.22**	−74°	55 % [$6b_g \rightarrow 12a_u$], 12 % [$5a_u \rightarrow 10b_g$]
					16^1B_u	59.9	0.06	+50°	59 % [$4b_g \rightarrow 9a_u$], 11 % [$4b_g \rightarrow 8a_u$]
					17^1B_u	60.5	**0.38**	−64°	49 % [$2b_g \rightarrow 6a_u$], 13 % [$3a_u \rightarrow 7b_g$]
					18^1B_u	61.5	**0.10**	+79°	37 % [$3b_g \rightarrow 7a_u$], 16 % [$3b_g \rightarrow 8a_u$]
					19^1B_u	61.7	**0.11**	−73°	15 % [$4a_u \rightarrow 9b_{ug}$], 10 % [$6b_g \rightarrow 13a_u$]
					20^1B_u	62.3	0.00	+19°	22 % [$6b_g \rightarrow 13a_u$], 15 % [$4a_u \rightarrow 9b_g$]

[a] Peak wavenumber in 1000 cm^{-1}.
[b] $3E_{ISO} = E_U + 2E_V$ (Fig. 1, top).
[c] Orientation factor (Section 4.1).
[d] In-plane transition moment angle (Scheme 2).
[e] Selected terms. Full listing provided as S2.
[f] Oscillator strength.
[g] Some important MOs are shown in Fig. 3.

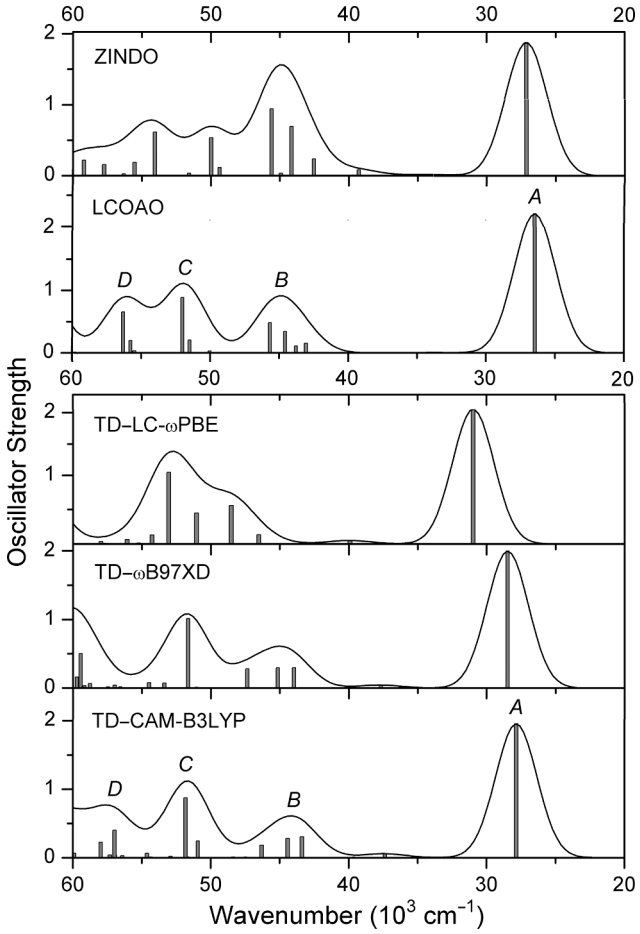


Fig. 2. Gaussian convolutions of electronic transitions for 1,4-distyrylbenzene (DSB) predicted with the semiempirical all-valence-electrons methods ZINDO and LCOAO, and with TD-DFT procedures using the functionals LC-ωPBE, ωB97XD, and CAM-B3LYP.

featureless. The absorption in this spectral region is likely to contain overlapping contributions from several electronic transitions. *K* s around 0.9 and 0.1 are estimated for the two maxima. These values are necessarily approximate, but they clearly indicate that the bands *C* and *D* are dominated by differently polarized contributions.

Further analysis of the LD data is complicated by the low molecular symmetry of DSB. We shall assume that the observed absorption is primarily due to π-π* transitions and thus polarized in the molecular plane (this assumption is supported by the calculated results (S2)). But according to the C_{2h} molecular point group, infinite moment directions are possible for in-plane polarized transitions. We thus need to determine the angles ϕ_i formed by the moments of the observed transitions *i* with a specific, well-defined axis in the plane. This axis is usually chosen as the "orientation axis" [20,21], that is the molecular axis *y* corresponding to the largest value of the average cosine squared, $\langle\cos^2(y,U)\rangle = K_y$, also called the "long axis" of the molecule. The in-plane "short axis" *x* perpendicular to *y* corresponds to the lowest average cosine squared among directions in the plane, $\langle\cos^2(x,U)\rangle = K_x$ [20,21]. The C_2 symmetry axis *z* is perpendicular to *x* and *y*, and we have $K_x + K_y + K_z = 1$. For a transition *i* polarized in the molecular *x,y* plane, the following relation holds [20]:

$$|\phi_i| = \tan^{-1}\sqrt{(K_y - K_i)/(K_i - K_x)} \tag{3}$$

Provided the orientation factors K_x and K_y for the in-plane axes *x* and *y* can be derived, the numerical values of the individual transition moment angles ϕ_i can thus be estimated from the observed K_i values. In

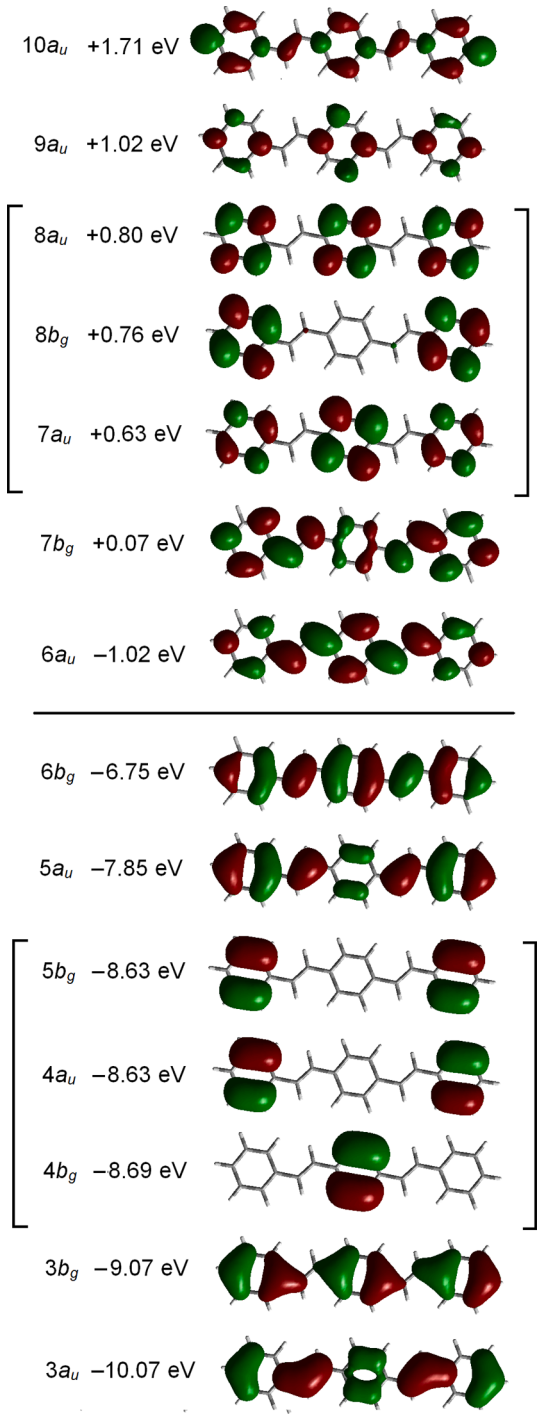
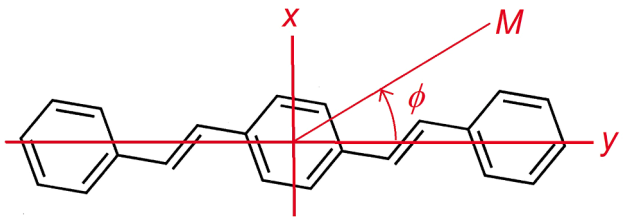


Fig. 3. Surface diagrams of the seven highest occupied and seven lowest unoccupied π-type MOs for 1,4-distyrylbenzene (DSB) computed with CAM-B3LYP/AUG-cc-pVTZ with indication of symmetries and energies.

view of the highly elongated molecular shape of DSB we shall assume that the orientation in stretched PE is rod-like [20], $K_y \gg K_x = K_z$, with orientation axis *y* corresponding to the longest molecular dimension (see Scheme 2). According to the calculated results (see below), the transition responsible for the intense band *A* is polarized along this axis (S2), and we shall thus adopt the value $K_y = K_A = 0.90$ (the corresponding Saupe orientation parameter [23,24] is $S = (3K-1)/2 = 0.85$). The absolute moment angles $|\phi_i|$ derived from Eqn. (3) with $K_y = 0.90$ and $K_x = (1-K_y)/2 = 0.05$ are listed in Table 1.



Scheme 2. Definition of in-plane transition moment angle ϕ (see text).

4.2. Electronic transitions

Roldao et al. [17] have recently compared the electronic transitions of DSB obtained with several advanced theoretical procedures, including CASPT2//CASSCF, considering the near UV region up to about 40,000 cm^{-1} (250 nm). The orbital structure of DSB is characterized by near-degeneracies of π and π* MOs in the regions next to the frontier region (Fig. 3, S2, S3), just as in the case of BPEB [19] which is *iso*-π-electronic with DSB. The predicted states in the high-wavenumber range thus tend to be of a composite nature, with essentially first order interactions between several near-degenerate electronic configurations, and the computed transitions are likely to be sensitive to calculational details. The transitions obtained with the present collection of semi-empirical and TD-DFT theoretical procedures are compared in Fig. 2 covering the region up to 60,000 cm^{-1} (167 nm). The overall structure of the observed spectrum with four main band systems *A*, *B*, *C*, and *D* (Fig. 1) is not well-reproduced by the TD–LC-ωPBE calculation, and ZINDO fails to predict the observed intensity distribution. The TD–ωB97XD and TD–CAMB3LYP results are quite similar, particularly in the regions of band *A* and *B*. We shall base our discussion on the TD–CAM-B3LYP (Table 1, S2) and LCOAO (S3, S4) results which are mutually consistent (Fig. 2) and in agreement with the observed general band structure throughout the investigated range.

The intense band *A* has an onset at 26,200 cm^{-1} (382 nm) and a maximum at 27,600 cm^{-1} (362 nm). Relative to spectra of DSB in liquid solution [4,6,17,40,41], the resolution of the vibrational fine structure of band *A* is improved in the present spectrum. This can probably be explained by the planarization effect of the stretched PE medium [19,55–57], reducing the torsional mobility of the solute molecule. Band *A* must be assigned to the 1^1B_u state predicted at 27,800 cm^{-1} (360 nm). This state is well described by the HOMO-LUMO excitation, $6b_g(\pi) \rightarrow 6a_u(\pi^*)$ (see Fig. 3) and is predicted to be polarized along the long axis *y* (for details, see S2). The corresponding long axis-polarized band in the spectrum of BPEB (Scheme 1) was observed at 28,900 cm^{-1} (346 nm) [19].

Several transitions are predicted in the region between bands *A* and *B* (Table 1), but they are not clearly observed in the present investigation. Transitions to the optically allowed states 2^1B_u and 3^1B_u computed at 37,400 and 39,600 cm^{-1} (267 and 253 nm) are weak. They are essentially long axis-polarized and are likely to be buried under the tail of the strong, long axis-polarized band *A*. In the case of BPEB (Scheme 1) with D_{2h} molecular symmetry, transitions to the corresponding states 1^1B_{2u} and 2^1B_{2u} are short axis-polarized and a weak feature at 37,400 cm^{-1} (267 nm) in the LD spectrum of BPEB was assigned to these transitions [19]. Based on an analysis of the fluorescence emission and excitation spectra of DSB, Gierschner et al. [4,5] estimated a wavenumber around 33,000 cm^{-1} (300 nm) for the 2^1B_u state. The state may possibly be observed directly by MCD spectroscopy, LCOAO predicts a positive MCD B-term for 1^1B_u but a negative B-term for 2^1B_u computed at 33,600 cm^{-1} (298 nm) with this method (S3, S4).

The states 2^1A_g, 3^1A_g, and 4^1A_g of DSB calculated at 36,000, 39,800, and 41,300 cm^{-1} (278, 251, and 242 nm) are electronically forbidden, but they may borrow long axis-polarized intensity by vibronic coupling with the strong band *A* via b_u symmetric molecular vibrations. The ^1A_g states are of particular interest in excited state absorption (ESA) spectroscopy; a strong feature at 1.67 eV (13,500 cm^{-1}) in the ESA spectrum of DSB was thus assigned to the transition $1^1B_u \rightarrow 3^1A_g$ by Oliveira et al. [14] (see also [17]).

Band *B* displays three closely spaced features at 39,700, 41,000, and 43,700 cm^{-1} (252, 244, and 229 nm). This absorption band is most likely due to the electronic states 4^1B_u, 5^1B_u, and 6^1B_u predicted at 43,400, 44,400, and 46,300 cm^{-1} (230, 225, and 216 nm) (Table 1), but the strong overlap between the observed features prevents a more detailed assignment. The estimated experimental moment angles $|\phi|$ (Table 1) indicate directions in the intermediate range between short and long axis polarization, consistent with the fluorescence anisotropy reported for a tetra-*t*-butyl derivative of DSB [3]. The theoretically predicted moment angles +24°, +61°, and +45° (Table 1) are in qualitative agreement with the experimental estimate. Strong MCD B-terms with alternating signs are predicted in the region of band *B* (S3, S4), an analysis of the MCD spectrum may be rewarding.

The high wavenumber region is dominated by two strong bands *C* and *D* with maxima near 49,800 and 57,500 cm^{-1} (201 and 174 nm) and absolute moment angles $|\phi|$ close to 0° and 76° (Table 1). The estimated angles are approximate (Section 4.1), but they clearly indicate that the bands *C* and *D* are predominantly long and short axis-polarized, respectively. The theoretical prediction of the numerous electronic states expected in this region is difficult, and the results of the present calculations must be considered as preliminary. We tentatively assign band *C* to the states 8^1B_u and 9^1B_u predicted at 50,900 and 51,800 cm^{-1} (196 and 193 nm) with moment angles ϕ equal to −8° and −7° (Table 1). A number of states are predicted to contribute to band *D*, primarily 14^1B_u and 15^1B_u at 57,000 and 58,000 cm^{-1} (175 and 172 nm) with ϕ equal to + 58° and −74°, but several additional states are likely to be involved. Within the limits of significance, we find that the transitions predicted with TD–CAM-B3LYP/AUG-cc-pVTZ are in pleasing agreement with the observed evidence (Table 1). Also the LCOAO results are consistent with the experimental evidence, even in the region of high wavenumbers (S3).

5. Conclusions

In this work, LD polarization data in the range 15,000–58,000 cm^{-1} (667–172 nm) are provided for DSB dissolved and aligned in stretched PE, leading to the characterization of four main band systems *A*, *B*, *C*, and *D* with maxima at 27,600, 41,000, 49,800, and close to 57,500 cm^{-1} (362, 244, 201, and 174 nm). The intense band *A* has been the object of several previous investigations, particularly in connection with studies of the photophysical properties of DSB [1–18]. The bands at higher wavenumbers seem to have attracted less attention. The analysis of the LD spectra is complicated by the low molecular symmetry of DSB, compared, for example, with the related chromophore BPEB (Scheme 1) [19]. According to the present results, bands *A* and *C* can be characterized as long axis-polarized, band *D* as predominantly short axis-polarized, while the features of band *B* are polarized at intermediate angles. We find that the theoretical transitions predicted with TD–CAM-B3LYP [39] and LCOAO [30,31] are in general agreement with the observed bands and their polarization directions. The ZINDO [28,29] and TD–LCωPBE [36,37] procedures are less successful.

Author contributions

Duy Duc Nguyen conceived and designed the experiments, performed the experiments, analyzed the data, authored or reviewed drafts of the paper, approved the final draft. *Nykola C. Jones* and *Søren V. Hoffmann* conceived and designed the experiments, performed the experiments, contributed reagents/materials/analysis tools, authored or reviewed drafts of the paper, approved the final draft. *Jens Spanget-Larsen* conceived and designed the experiments, analyzed the data, contributed reagents/materials/analysis tools, prepared figures and/or tables, performed the computation work, authored or reviewed drafts of the paper,

approved the final draft.

Declaration of Competing Interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

Data availability

Spectroscopic data are available from the UV-Vis$^+$ Photochemistry Database (https://science-softcon.de/spectra/).

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Appendix A. Supplementary material

Supplementary data to this article can be found online at https://doi.org/10.1016/j.saa.2022.122019.

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